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Original research paper

4 **Total phenolic content, flavonoid content and antioxidant potential of *Petasites hybridus***
5 **and related species from Croatia and considerations regarding their**
6 **pharmaceutical significance**

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ABSTRACT

26 Extracts obtained from common butterbur (*Petasites hybridus*), standardized to petasins, are
27 existing pharmaceutical options for allergic rhinitis (leaves) and migraine (rhizomes) treatment
28 and/or prevention. In this study, total phenolic content, flavonoid content and antioxidant po-
29 tential of ten samples of Croatian *Petasites* species (four *P. hybridus*, four *P. albus*, one *P.*
30 *kablikianus*, and one *P. paradoxus*) obtained by ultrasound-assisted extraction of leaves were
31 compared. The total phenolic content (Folin-Ciocalteu assay) of methanolic leaf extracts ranged
32 from 4.43 ± 0.09 to 10.76 ± 0.60 mg gallic acid equivalent g^{-1} dry weight (mg GAE g^{-1} DW)
33 for *P. hybridus* and from 6.66 ± 0.43 to 19.92 ± 2.90 mg GAE g^{-1} DW for *P. albus* samples,
34 while those of *P. kablikianus* and *P. paradoxus* were equal to 7.56 ± 0.17 mg GAE g^{-1} DW and
35 10.22 ± 0.46 mg GAE g^{-1} DW, respectively. Flavonoid content (AlCl_3 assay) varied between
36 2.51 ± 0.10 and 4.03 ± 0.08 mg quercetin equivalent g^{-1} dry weight (mg QE g^{-1} DW) for *P.*
37 *hybridus* and between 2.21 ± 0.09 and 5.22 ± 0.02 mg QE g^{-1} DW for *P. albus* samples, while
38 those of *P. kablikianus* and *P. paradoxus* were equal to 5.59 ± 0.05 mg QE g^{-1} DW and $5.50 \pm$
39 0.09 mg QE g^{-1} DW, respectively. Antioxidant potential was in high correlation with total phe-
40 nolic content ($r = 0.93$, $p < 0.001$). Due to expected contribution of plant polyphenols and fla-
41 vonoids to the activity of butterbur extracts and their observed great variabilities, determining
42 the content of these compounds may be of interest to the pharmaceutical industry.

43 *Keywords:* *Petasites*, Asteraceae, phytotherapy, green synthesis, polyphenols, DPPH assay

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INTRODUCTION

48 *Petasites* Mill. (butterbur) species are large-leaved rhizomatous perennial herbs from the
49 daisy (Asteraceae) family, which are distributed all over Europe, Asia and North America. Their

50 name comes from the Greek word *petasos*, which refers to a large hat commonly worn in An-
51 cient Greece. Out of 19 species that are accepted today (1), the widespread *Petasites hybridus*
52 G.Gaertn., B.Mey. & Scherb. (common butterbur) is pharmaceutically and phytotherapeutically
53 the most important (2). In recent years, the use of its extracts as green media for generating
54 magnetic organometallic nanocomposites that are used as catalysts in one-pot multicomponent
55 reactions (MCRs) is becoming increasingly popular, *e.g.*, for the synthesis of compounds that
56 could reduce organic pollutants or that could be used in synthesis of new pharmaceuticals (3,
57 4). Flavonoids and phenolics present in the *P. hybridus* rhizome extract were suggested as the
58 main compounds responsible for the reduction of metal ions to nano zero-valent metallic parti-
59 cles (3). Modern pharmacotherapy also recognizes the use of *P. hybridus* rhizome extract
60 (Petadolex[®]) for the prevention of migraine attacks (5), while *P. hybridus* leaf extract (Tesalin[®]
61 – Ze 339) is clinically approved as a herbal medicine for the treatment of symptoms of seasonal
62 allergic rhinitis such as rhinorrhea, sneezing, and nasal congestion (1). In fact, out of twenty-
63 nine randomized controlled trials which evaluated the use of single medicinal plants for allergic
64 rhinitis among adults and children, the greatest number of studies included *P. hybridus* (6).

65 Rare and idiosyncratic cases of herb-induced liver injury caused by the rhizome extract
66 Petadolex[®], often confounded by hepatotoxic co-medications, were reported (N = 48 cases over
67 a period of > 30 years and an estimated 2.6 million patient month exposure), while clinically
68 relevant liver function abnormalities were not observed in clinical trials with migraine patients
69 (5). On the other hand, there have been no reports of serious adverse drug reactions with the
70 butterbur leaf extract so far (6). This was recently acknowledged by the Swiss health authority
71 as Tesalin[®] (Ze 339) was switched from prescription to the nonprescription status. It was con-
72 cluded that *P. hybridus* leaf extract Ze 339 may be regarded as safe if used in the clinically
73 recommended dose regime (7). Also, results from a recent study that evaluated *in vivo* single
74 and repeated oral dose toxicity and *in vitro* genotoxicity of *P. japonicus* (Siebold & Zucc.)
75 Maxim. leaves, suggested that they may be safe for human consumption (8). The clinically

76 approved butterbur extracts mentioned above are standardized to petasins and are declared as
77 PA-free, *i.e.*, free of hepatotoxic and carcinogenic pyrrolizidine alkaloids (PAs). Due to their
78 initially lower contents of PAs, leaves may be a more suitable source of petasins (9), the phar-
79 macologically active ingredients at least partially responsible for the anti-inflammatory effects
80 of butterbur extracts (5). Various populations of *P. hybridus* were shown to vary considerably
81 in petasin content both in their rhizomes (7.4 to 15.3 mg g⁻¹ dry weight (DW)) and leaves (3.3
82 to 11.4 mg g⁻¹ DW), while even greater differences were observed between rhizomes (4.8 –
83 89.9 µg g⁻¹ DW) and leaves (0.02–1.50 µg g⁻¹ DW) in the content of PAs (9). Besides between
84 different organs, PA content may vary considerably within and between populations, while sea-
85 sonal variations seem to be of minor importance (10). Similarly, great variabilities in essential
86 oil constituents were observed for different plant parts and populations of *P. hybridus* and *P.*
87 *albus* (L.) Gaertn. from Croatia (11). Recently, also the content of total phenolic compounds,
88 antioxidant and antimicrobial activity were reported for extracts obtained by ultrasound-as-
89 sisted extraction (UAE) of different plant parts of *P. hybridus* from Turkey (12). However, little
90 is known about the possible variabilities of phenolics and flavonoids between different popula-
91 tions of *P. hybridus* and related species.

92 Besides *P. hybridus*, which has been recognized as one of the most important Central Eu-
93 ropean medicinal herbs used from classical antiquity to modern and contemporary era (13),
94 other species from the genus *Petasites* such as *P. japonicus*, *P. tricholobus* Franch., *P. for-*
95 *mosanus* Kitam., and *P. frigidus* (L.) Fries have been used worldwide both as food and tradi-
96 tional medicines (1, 14, 15). In Bosnia and Herzegovina, ointments prepared from leaves of
97 wild *P. hybridus* and *P. albus* are used for rheumatism (16). Ethnomedicinal use of the same
98 two species was recently reported from Serbia (17). The aim of this study was to evaluate the
99 total phenolic and flavonoid contents of *P. hybridus*, *P. albus*, *P. kablikianus* Bercht., and *P.*
100 *paradoxus* (Retz.) Baumg. together with their antioxidant potential. Samples of the former two
101 species were collected from four different locations in Croatia, two of which were shared by

102 both species (Medvednica, Fužine). To our knowledge, this is the first study that compared the
103 phytochemical content and antioxidant activity of several *Petasites* species, and the first such
104 study that compared the contents of polyphenols and flavonoids in different populations of *P.*
105 *hybridus* and *P. albus*.

106

107

EXPERIMENTAL

108 *Plant material*

109 Leaves of four different *Petasites* species were collected during the flowering period from
110 ten wild populations in Croatia: *Petasites hybridus* (Mount Medvednica, Fužine, Mount
111 Ivanščica, Zagreb - Maksimir), *P. albus* (Mount Medvednica, Fužine, Mount Risnjak, Northern
112 Velebit), *P. kablikianus* (Plitvice Lakes), and *P. paradoxus* (Baške Oštarije). Samples were
113 authenticated by Prof. Kroata Hazler Pilepić and voucher specimens deposited at the herbarium
114 of the Department of Pharmaceutical Botany, University of Zagreb Faculty of Pharmacy and
115 Biochemistry.

116

117 *Extract preparation*

118 After air-drying at room temperature, the leaves were initially cut into smaller pieces and
119 then ground to powder using an electric mill. Ultrasound-assisted extraction (UAE) was per-
120 formed twice by adding 5 mL of methanol to 0.5 g of powdered plant material in the duration
121 of 2 × 30 minutes. After filtering, the two extracts were combined and made up to the mark
122 with methanol in a 10 mL flask.

123 *Chemical reagents and standards*

124 Folin-Ciocalteu's reagent and sodium carbonate decahydrate were purchased from Kemika
125 (Croatia). Aluminum chloride hexahydrate was obtained from Sigma-Aldrich (USA) and 2,2-
126 diphenyl-1-picrylhydrazyl (DPPH) from Fluka (Switzerland). Standard compounds gallic acid,

127 quercetin and 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox), were pur-
128 chased from Merck Schuchardt (Germany), Sigma-Aldrich (USA) and Acros Organics (Bel-
129 gium), respectively. All chemicals used were of analytical grade.

130 *Evaluation of total phenolic and flavonoid content*

131 Total phenolic content was evaluated using the Folin-Ciocalteu's reagent, similarly as pre-
132 viously described (18). In brief, 0.25 mL of extract was mixed with 1.25 mL of Folin-Ciocal-
133 teu's reagent (10 % (v/v)). After 5 min, 1 mL sodium carbonate decahydrate (7.5 g 100 mL⁻¹)
134 was added. After 60 min of incubation, the absorbance was read at 765 nm. The results are
135 expressed as mg gallic acid equivalent g⁻¹ dry weight (mg GAE g⁻¹ DW).

136 Flavonoid content was evaluated using the previously described aluminum chloride method
137 (18). In brief, 1 mL of extract was mixed with 1 mL AlCl₃ × 6 H₂O (2 g 100 mL⁻¹). After 15
138 min, absorbance was measured at 415 nm. The results are expressed as mg quercetin equivalent
139 g⁻¹ dry weight (mg QE g⁻¹ DW).

140 *DPPH assay*

141 Antioxidant potential was evaluated based on the DPPH radical scavenging activity as pre-
142 viously described (18). To 2 mL of methanolic DPPH solution, adjusted to initial absorbance
143 of 0.70 ± 0.02, 10 µL of extract was added. After 30 min incubation, the decrease in absorption
144 of the radical was measured at 517 nm. The results are expressed as mg Trolox equivalent g⁻¹
145 dry weight (mg TE g⁻¹ DW).

146

147 *Statistical analysis*

148 All measurements were performed in triplicate. The results are expressed as means ± stand-
149 ard deviations (SD). Correlations between measured parameters were assessed using Pearson's
150 correlation coefficient (*r*) with the significance level, *α*, set at 0.05. Statistical analysis was
151 performed in GraphPad Prism 9.0. (GraphPad Software, San Diego, USA).

152

154 In this study, four European *Petasites* species were harvested from ten wild populations in
155 Croatia in order to evaluate their (poly)phenolic content, flavonoid content and antioxidant po-
156 tential: *P. hybridus* (Medvednica, Fužine, Ivanščica, Zagreb - Maksimir), *P. albus*
157 (Medvednica, Fužine, Risnjak, Northern Velebit), *P. kablikianus* (Plitvice Lakes), and *P. par-*
158 *adoxus* (Baške Oštarije).

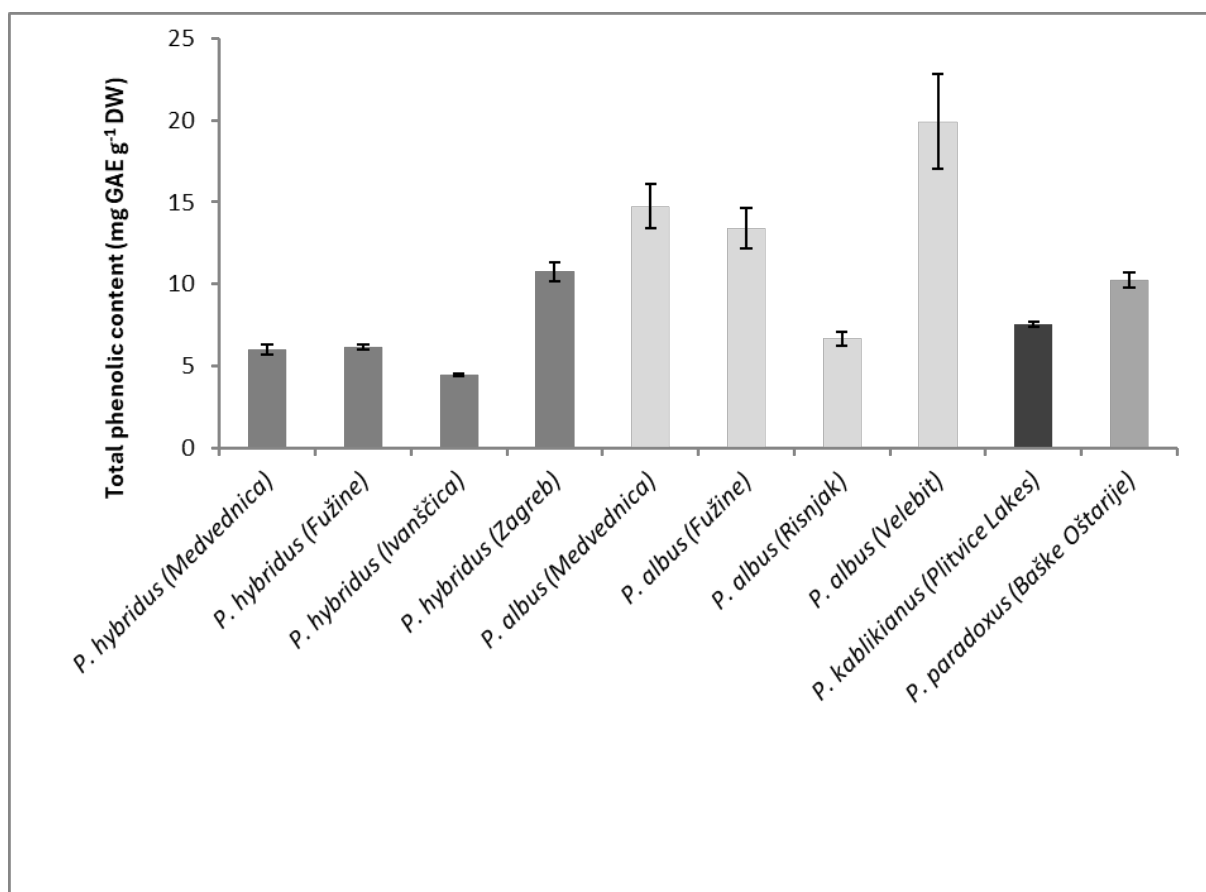
159 *Total phenolic content*

160 The total phenolic content of collected samples was assessed using the Folin-Ciocalteu as-
161 say, one of the most commonly used methods for the determination of (poly)phenolic com-
162 pounds in plant-based foods and beverages (19). The assay is based on a single electron-transfer
163 (SET) in which the antioxidant species acts as the electron donor and the Folin–Ciocalteu’s
164 reagent acts as the oxidant, causing a change in color from yellow to blue, directly proportional
165 to the reducing activity of the phenolic compounds. This is frequently displayed as gallic acid
166 equivalents (GAE) (20).

167 In the present study, the total phenolic content of *P. hybridus* samples ranged from $4.43 \pm$
168 0.09 to 10.76 ± 0.60 mg GAE g⁻¹ DW, while for *P. albus* samples it varied between 6.66 ± 0.43
169 and 19.92 ± 2.90 mg GAE g⁻¹ DW (Fig. 1). Furthermore, the total phenolic contents of the
170 remaining two species were within the ranges observed for *P. hybridus* and *P. albus*, $7.56 \pm$
171 0.17 mg GAE g⁻¹ DW in *P. kablikianus* and 10.22 ± 0.46 mg GAE g⁻¹ DW *P. paradoxus*. In a
172 recent study from Turkey, the content of total phenolic compounds of *P. hybridus* leaf extract,
173 under optimal conditions, was found to be 3.78 µg GAE mg⁻¹ extract (12). Total phenolic acid
174 content of extracts obtained from *P. japonicus* leaves and stalks, assessed from peak areas of
175 the UPLC-DAD chromatogram, was 16.76 ± 0.42 mg g⁻¹ DW. The major phenolic acid was
176 3,5-di-*O*-caffeoylquinic acid followed by 5-*O*-caffeoylquinic acid and fukinolic acid, while
177 kaempferol 3-*O*-(6''-*O*-acetyl) glucoside, quercetin 3-*O*-(6''-*O*-acetyl) glucoside, astragalín,

178 and kaempferol 3-*O*-rutinoside (nicotiflorin) were the most represented flavonoids (21). The
179 presence of caffeoylquinic and feruloylquinic acid derivatives was reported in *P. hybridus*
180 leaves as well (22).

181



182

183 Fig. 1. Total phenolic content of ten investigated *Petasites* samples (averages \pm SD, $n = 3$),
184 GAE – gallic acid equivalent, DW – dry weight.

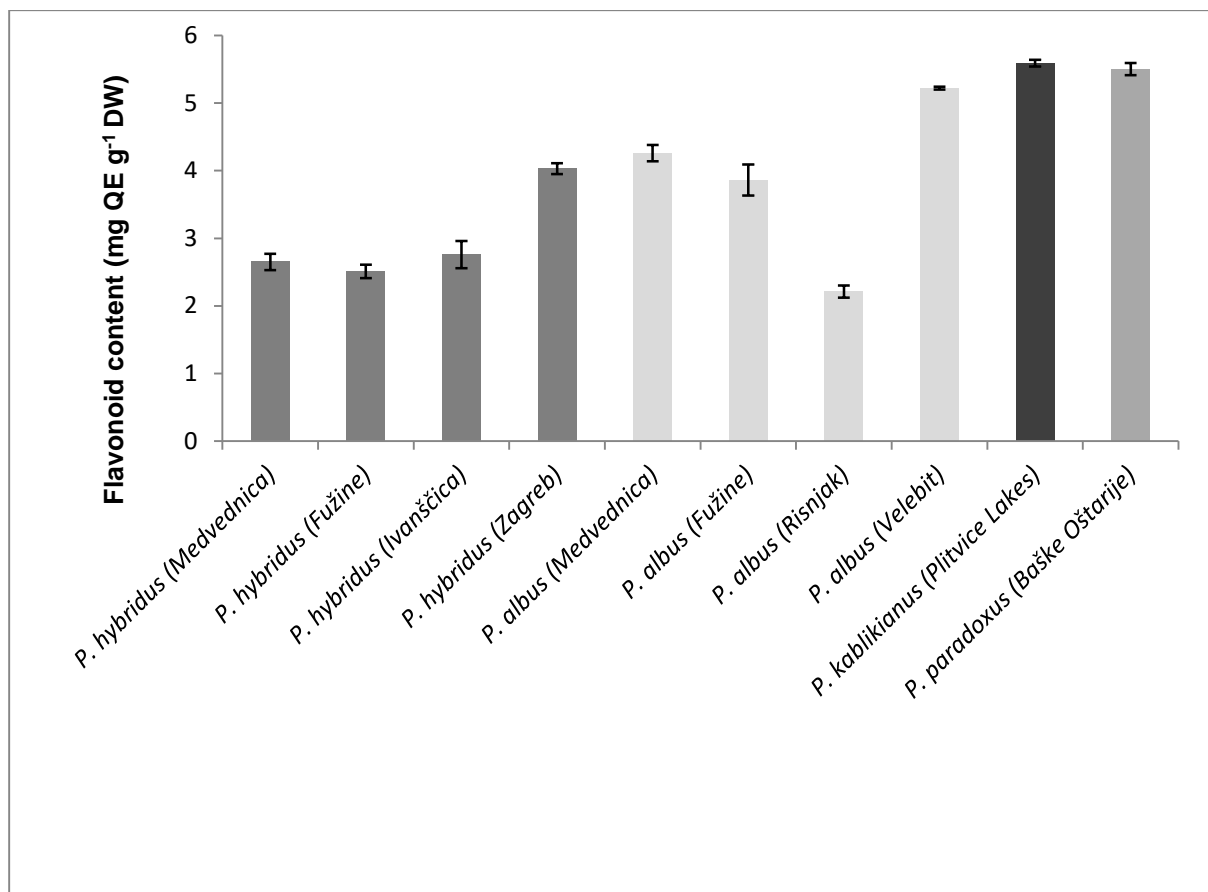
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186 *Flavonoid content*

187 Flavonoid content in this study was assessed based on a spectrophotometric aluminum chlo-
188 ride chelation method, one of the most commonly used methods for the so-called total flavonoid
189 determination, using the procedure without NaNO₂ (23), which may only be used for the esti-
190 mation of the contents of certain classes of flavonoids, *i.e.*, flavones and flavonols (24). Flavo-
191 nols such as quercetin and kaempferol and/or their glycosides were previously reported for

192 *Petasites* species (21, 25). In the assay, flavonols form complexes with Al(III) with C-3 and C-
193 5 hydroxy groups and with the dihydroxy groups in B ring (23).

194 The highest flavonoid contents were recorded for *P. kablikianus* and *P. paradoxus*, $5.59 \pm$
195 $0.05 \text{ mg QE g}^{-1} \text{ DW}$ and $5.50 \pm 0.09 \text{ mg QE g}^{-1} \text{ DW}$, respectively (Fig. 2). For *P. hybridus*, the
196 same varied between 2.51 ± 0.10 and $4.03 \pm 0.08 \text{ mg QE g}^{-1} \text{ DW}$, while for *P. albus*, a wider
197 range of flavonoid contents was recorded (from 2.21 ± 0.09 to $5.22 \pm 0.02 \text{ mg QE g}^{-1} \text{ DW}$).
198 Not a lot is known about the flavonoids from investigated *Petasites* species. For *P. hybridus*,
199 quercetin, quercitrin and rutin were reported (1). Flavone glycosides quercetin-3-*O*- β -glucopy-
200 ranoside and kaempferol-3-*O*- β -glucopyranoside and their rhamnosylated derivatives (rutin,
201 nicotiflorin) were isolated from *P. tricholobus* (25). Recently, a more detailed list of flavonoid
202 compounds was given for 80 % ethanolic extract obtained by Soxhlet extraction from *P. japon-*
203 *icus* leaves and stalks, which was based on ultra performance liquid chromatography coupled
204 with diode array detector, quadrupole time-of-flight mass spectrometry (UPLC-QToF-MS)
205 characterization. The total flavonoid content of the same extract was estimated to be $10.65 \pm$
206 $1.25 \text{ mg g}^{-1} \text{ DW}$ (21).



207

208

209 Fig. 2. Flavonoid content of ten investigated *Petasites* samples (averages \pm SD, $n = 3$), QE
 210 – quercetin equivalent, DW – dry weight.

211

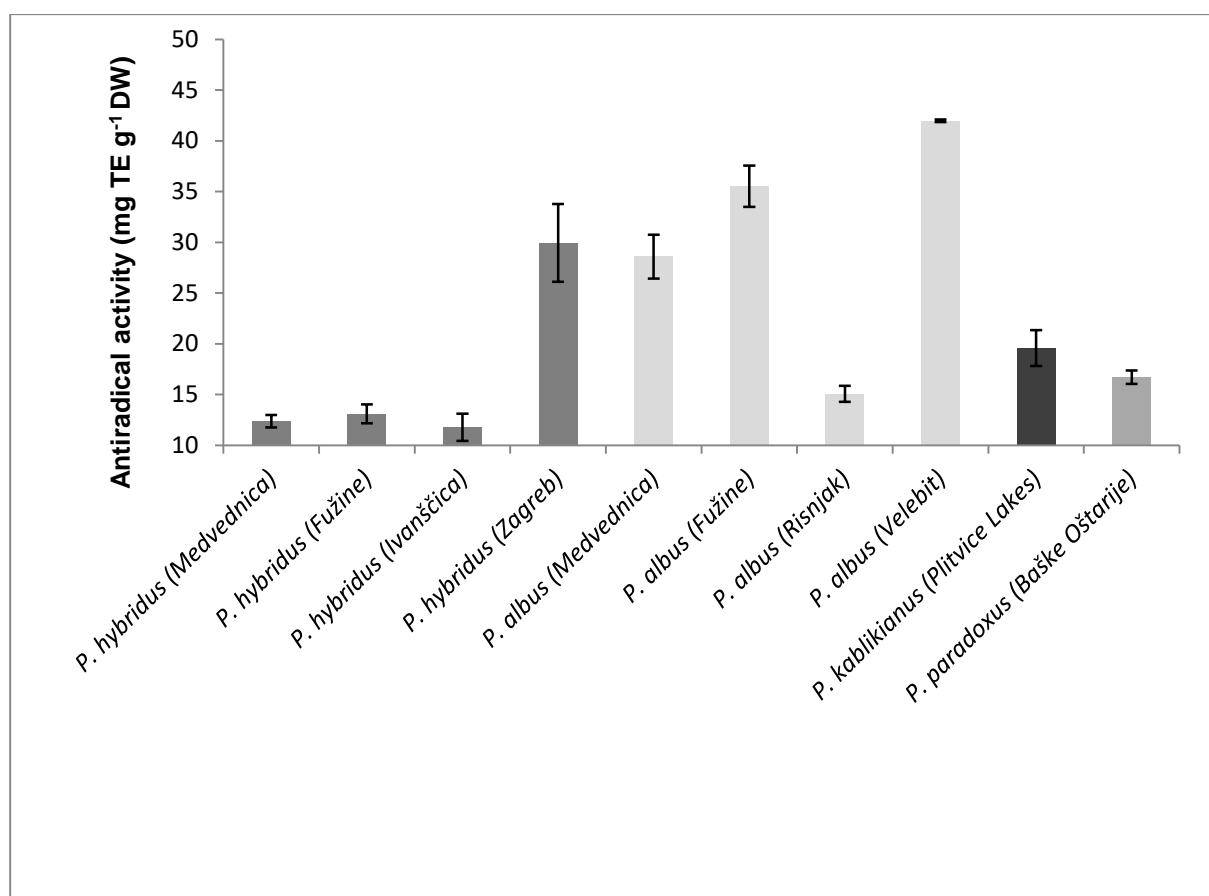
212 *Antioxidant potential*

213 Antioxidant potential was evaluated using the DPPH radical scavenging assay, the most
 214 known and commonly used method to determine antioxidant ability of food and pharmaceutical
 215 ingredients. The DPPH assay is based on spectrophotometric measurements of the capacity of
 216 antioxidants to scavenge DPPH radicals. The method is popular due to its ease of use, sensitiv-
 217 ity, reproducibility, and execution speed, and can be readily used without the need for free
 218 radical preparation prior to performing the test (26). The reaction mechanism is primarily based
 219 on the electron transfer (ET), which can be further subdivided into single electron transfer fol-
 220 lowed by a proton transfer (SET-PT) and sequential proton loss electron transfer (SPLET) (27).

221 Also, a marginal reaction pathway in the assay is hydrogen atom transfer (HAT), *i.e.*, reduction
222 of the single electron of the nitrogen atom in DPPH radical (intense deep violet color) to the
223 corresponding hydrazine DPPH-H (pale yellow color) by taking a hydrogen atom from the an-
224 tioxidants (H-donors) (28). A standard antioxidant Trolox may be used as a reference standard
225 and the results of the assay may be expressed as Trolox equivalents (TE) (26).

226

227



228

229 Fig. 3. DPPH radical scavenging activity of ten investigated *Petasites* samples (averages ±
230 SD, $n = 3$), TE – Trolox equivalent, DW – dry weight.

231

232 The DPPH radical scavenging activity found in this study ranged from 11.78 ± 1.34 to 29.94
233 ± 3.83 mg TE g⁻¹ DW in *P. hybridus* samples and from 15.07 ± 0.80 to 41.98 ± 0.12 mg TE

234 g^{-1} DW in *P. albus* samples, while it was within the same ranges for the remaining two species,
 235 *P. kablikianus* (19.59 ± 1.77 mg TE g^{-1} DW) and *P. paradoxus* (16.72 ± 0.66 mg TE g^{-1} DW)
 236 (Fig. 3). The antioxidant activity estimated by the DPPH assay of *P. hybridus* ethanolic leaf
 237 extract obtained under optimized conditions was found to be $2.27 \mu\text{g TE mg}^{-1}$ extract (12). In
 238 the present study, the DPPH radical scavenging activity was found to be in high correlation
 239 with the total phenolic content ($r = 0.93$, $p < 0.001$, Table I). This may be explained by the fact
 240 that both assays are based on SET reactions (20, 27).

241 *Table I. Pearson's correlation coefficients (r) between the measured parameters (total phe-*
 242 *nolic content, flavonoid content, and DPPH radical scavenging activity) of ten Petasites sam-*
 243 *ples (one significant correlation is marked in bold)*

244

Measured parameters	Total phenolic content	Flavonoid content
Flavonoid content	0.59 ($p = 0.071$)	1
DPPH radical scavenging activity	0.93 ($p < 0.001$)	0.52 ($p = 0.125$)

245 Until today, most of the investigations of antioxidant activity of *Petasites* species were done
246 on *P. japonicus* and the DPPH assay was the most frequently used *in vitro* method (14). Caffeic
247 acid derivatives such as 5-*O*-caffeoylquinic acid, fukinolic acid, 3,5-di-*O*-caffeoylquinic acid,
248 and 4,5-di-*O*-caffeoylquinic acid present in methanolic extracts of leaves and roots, were some
249 of the compounds responsible for the antioxidant activity with 3,5-di-*O*-caffeoylquinic acid
250 having the greatest radical scavenging capacity in leaf (23.09 %) and root extracts (26.47 %).
251 On the other hand, flavone glycoside quercetin-3-*O*-(6''-acetyl)- β -glucopyranoside, which was
252 present only in the leaf extract, showed weak activity, while no activity was observed for
253 kaempferol-3-*O*-(6''-acetyl)- β -glucopyranoside (29). This is consistent with our results, ac-
254 cording to which the antiradical activity of investigated *Petasites* species was not significantly
255 correlated with flavonoid content. Indeed, the aforementioned caffeoylquinic acid derivatives
256 were, likewise, reported to be present in *P. hybridus* leaves together with 5-*O*-feruloylquinic
257 acid, 1-*O*-caffeoyl-3-*O*-feruloylquinic acid, and 1-*O*-caffeoyl-4-*O*-feruloylquinic acid. It is in-
258 teresting to notice that other well-known medicinal plants such as *Achillea millefolium* L. (yar-
259 row) and *Cynara scolymus* L. (artichoke) also possessed most of the aforementioned hy-
260 droxycinnamic acid derivatives (22). Fukinolic acid is yet another phenolic compound, which
261 was recognized as the major antioxidant constituent in ethanolic extracts of *P. japonicus* flower
262 buds (30). However, to our knowledge, its presence in other *Petasites* species has not been
263 reported so far (31). On the other hand, in investigated plants, other non-phenolic antioxidants
264 may be present such as the benzofuran derivative euparin found in essential oils of *P. albus*
265 (aerial parts), which was observed to possess moderate antioxidant activity according to the
266 DPPH radical scavenging assay (32).

267 Evaluation of antioxidant activity of extracts prepared from leaves, stems and roots of *P.*
268 *japonicus* based on various methods, including DPPH, 2,2'-azinobis(3-ethylbenzothiazoline-6-
269 sulfonic acid) radical cation (ABTS⁺), and superoxide anion radical scavenging assay, and the
270 ferric reducing ability of plasma (FRAP), showed that the antioxidant activity of leaf extract is

271 superior to those of other extracts (33). Similarly, essential oils obtained from leaves of *P. hy-*
272 *bridus* subsp. *ochroleucus* exhibited greater antioxidant activity in the DPPH assay in compar-
273 ison to essential oils obtained from rhizomes of the same plant (34).

274 *Variabilities in specialized metabolites content within and between species*

275 As observed in this and in some of our previous studies, the contents of specialized (sec-
276 ondary) metabolites may not only differ between different species of the same genus (35), but
277 they may also vary considerably between different populations of the same species (18, 36).
278 For this reason, to gain a better insight into the richness of biologically active compounds of a
279 certain species, it is beneficial to include, when possible, more than one population of the spe-
280 cies of interest in the study. This is certainly more challenging in terms of sample collection
281 and is not always possible, *e.g.*, when the species of interest are growing in single and/or hard-
282 to-reach locations. Generally, investigated *Petasites* species are known to inhabit moist, damp
283 and shady areas, and are often covering large surfaces on riverbanks, near lakes and streams
284 thanks to their creeping underground stems (rhizomes) (1, 37). Common butterbur (*P. hybridus*)
285 and white butterbur (*P. albus*) are widespread in Europe and in Croatia (1, 38). On the other
286 hand, in Croatia, *P. kablikianus* (glabrous butterbur) is found only in the Plitvice Lakes area
287 (38). In fact, it is the dominant fast-decomposing species growing on the tufa barriers, whose
288 formation is promoted by leaf litter breakdown (and *vice versa*) (39). *Petasites paradoxus* (Al-
289 pine butterbur), on the other hand, as its names suggests, may be found in mountainous regions
290 (37, 38).

291 In this study, *P. kablikianus* and *P. paradoxus* were each collected from a single location,
292 while four different populations of *P. hybridus* and *P. albus* were investigated due to the better
293 distribution and accessibility of these species. Conveniently, two of the four harvesting loca-
294 tions were shared by the same species, which enabled a more reliable comparison between
295 them, bearing in mind that they had been growing in equal environmental conditions. In our

296 previous studies, contents of secondary metabolites in leaf extracts were observed to be corre-
297 lated with monthly precipitation amounts and mean monthly temperature (11, 18). Based on the
298 comparison of the results obtained for the samples harvested from the shared locations
299 (Medvednica and Fužine), it could be observed that *P. albus* contained more (poly)phenolic
300 compounds and more flavonoids than *P. hybridus*. The total phenolic content of these popula-
301 tions of *P. albus* was also higher than those of *P. kablikianus* and *P. paradoxus*. On the other
302 hand, the latter two species contained the most flavonoids. These results make the three species
303 of *Petasites* potentially interesting in terms of their possible exploitation as sources of natural
304 antioxidants and/or utilization as media in green synthesis of nanoparticles.

305 *Considerations regarding the use of Petasites species in green synthesis*

306 Green synthesis is a subdivision of green chemistry, which aims to develop safer and more
307 sustainable chemical products and procedures. The fundamental green chemistry principles that
308 are applied in green synthesis include environmental pollution mitigation, renewable feedstock
309 usage, usage of non-toxic (or safer) solvents/auxiliaries, derivatives usage minimization, and
310 waste prevention or reduction (40). Green synthesis of nanoparticles from biomass and waste,
311 as an eco-friendly, biocompatible, and cost-effective approach for use in medicine, agriculture,
312 environmental remediation, and other fields, is believed to allow up to 30 % reduction in energy
313 consumption, up to 40 % cost savings, and up to 50 % production increase and could therefore,
314 contribute to a more sustainable future (41). Compared to other green synthesis methods of
315 nanoparticles, plant-mediated synthesis is the most efficient (40).

316 Lately, the number of papers suggesting the use of *P. hybridus* rhizomes in green synthesis
317 is on the rise (3, 4). The root extract of this plant is considered a renewable, mild, and safe
318 reducing agent and effective stabilizer (4). Moreover, utilization of *P. hybridus* leaf extract has
319 also been demonstrated for the same purpose (42). However, caution should be taken consider-

320 ing that *Petasites* species are natural sources of highly hepatotoxic and carcinogenic pyrroliz-
321 idine alkaloids (PAs), especially in their underground parts (9, 43). Although the right mecha-
322 nism by which butterbur preparations might cause liver injury is not known, it is suggested that
323 liver related adverse effects of commercially available butterbur products were likely connected
324 to PAs contamination or mislabelling of the products (44). With that in mind, although the use
325 of these extracts may be possibly advantageous, it may also be potentially harmful, even when
326 low levels of PAs are present.

327 Leaves, which were used for *Petasites* extract preparation in this study, could potentially
328 serve as a better and more renewable source of biologically active compounds (*e.g.*, those acting
329 as reducing agents in the synthesis of nanoparticles/nanomaterials) in comparison to roots/rhi-
330 zomes, considering their expected higher content of reducing substances/antioxidants (33) and
331 significantly lower PAs content (9, 43). Leaves of other species such as *Eucalyptus* sp., *Thymus*
332 *vulgaris* L., and *Ginkgo biloba* L. have been successfully used to synthesize nanoparticles.
333 However, the main research focus should be put on materials that are not limited by seasonal
334 and geographical availability (45).

335 Utilization of leaves from *Petasites* species could be potentially interesting due to the world-
336 wide distribution of these species and their large size (1). The results of our study, although
337 based on simple spectrophotometric reactions and a relatively small number of samples, indi-
338 cate that *P. albus*, *P. kablikianus* and *P. paradoxus* may exhibit similar, if not better, antioxi-
339 dant/reducing properties to those of *P. hybridus*. Considering the need to minimize the risk from
340 exposure to PAs, it would be interesting to compare pyrrolizidine alkaloid contents of *P. hy-*
341 *bridus* and related species that may be locally available in future studies. Also, since infor-
342 mation on the phytochemical composition/constituents, especially those including phenolic ac-
343 ids, flavonoids, and other polyphenols of *P. albus*, *P. kablikianus* and *P. paradoxus* as well as
344 most other *Petasites* species are lacking or are relatively modest in the case of *P. hybridus*, it

345 would be interesting to evaluate those as well, having in mind their potential antioxidant (re-
346 ducing) properties important for green synthesis.

347

349 Up to now, five randomized controlled trials evaluated the use of *P. hybridus* standardized
350 leaf extract (Tesalin – Ze 339) for allergic rhinitis among adults and children and, to the best of
351 our knowledge, the extract has been used without reported serious side effects (6, 46). Also, in
352 a randomized, placebo-controlled trial, a fixed herbal drug combination composed of four plant
353 extracts including *P. hybridus* leaf extract (Ze 185) was recorded to be efficacious and safe in
354 short-term treatment of patients with somatoform disorders (47) and the same preparation was
355 observed to reduce self-reported anxiety response to stress in healthy men (48). The mentioned
356 extract Ze 339 is obtained by supercritical CO₂ fluid extraction (SFE-CO₂) and is standardized
357 to 8 mg petasins (petasin, isopetasin, and neopetasin) as active substances (49). Phenolic acids
358 and flavonoids have not been considered as important biologically active compounds present
359 in this extract.

360 Together with ultrasound-assisted extraction (UAE), pressurized liquid extraction, and mi-
361 crowave-assisted extraction, SFE-CO₂, is a green chemistry method used for extraction and
362 isolation of bioactive compounds from plants-based materials. It is considered as one of the
363 best techniques for obtaining flavonoids, essential oils, and other natural chemical components
364 from natural plant materials (50). Flower extracts obtained by SFE-CO₂, which have shown
365 anti-inflammatory activity, were often characterized by flavonoids (*e.g.*, quercetin, kaempferol,
366 quercetin 3-*O*-rhamnoside, quercetin 3-*O*-glucoside, quercetin 3-*O*-rutinoside) and phenolic ac-
367 ids as their major products (51). Similarly, a recent study on the flavonoid yield and profile of
368 *Ziziphus jujuba* leaves indicated that, compared with conventional Soxhlet extraction and UAE,
369 SFE-CO₂ with ethanol as a cosolvent may provide an extract with significantly increased fla-
370 vonoid yield, antioxidant activity and antiproliferative activity (52). The flavonoids compounds
371 identified in the study were kaempferol and quercetin glycosides including some that may be
372 found in *Petasites* species such as rutin and quercetin-3-*O*-glucoside.

373 It has been shown that the extract Ze 339 is five times as active as purified petasin indicating
374 that other constituents present in the plant material (extract matrix) may influence the biophar-
375 maceutical properties of active ingredients. However, variations of fatty acids (17.1 – 27.2 %),
376 crude oil and fat (17.7 – 44.2 %), sterols (3.0 – 4.9 %) and essential oils (1.3 – 10.5 %) observed
377 in a quantitative analysis of twelve extract batches did not result in significant differences in
378 inhibition of leukotriene synthesis (49). Polyphenolic compounds are important plant special-
379 ized metabolites, whose use may provide health promoting effects due to their diverse biologi-
380 cal activities including, but not limited to antioxidant, anti-inflammatory, and antiallergic ac-
381 tivity, making them interesting for industries such as food and pharmaceutical (53). It is possi-
382 ble that some of these compounds also contribute to the observed beneficial activities of but-
383 terbur and its preparations. Recently, *P. japonicus* leaf extract, which was standardized to 3,4-
384 dicaffeoylquinic acid ($0.02 \mu\text{g g}^{-1}$), 3,5-dicaffeoylquinic acid ($0.15 \mu\text{g g}^{-1}$), and 4,5-
385 dicaffeoylquinic acid ($0.43 \mu\text{g g}^{-1}$), showed neuroprotective activity against amyloid beta 25-
386 35 protein fragment ($A\beta_{25-35}$) plaque neurotoxicity *in vitro* and *in vivo* (54). The results of our
387 study indicate that antioxidant activity could be connected to the total phenolic content of these
388 extracts. Amounts of these compounds in *Petasites* samples harvested from different locations
389 may vary significantly, just as those of other biologically active substances, which in turn could
390 influence the overall activity of the extracts. Therefore, it may be interesting to analyze the
391 contents of these compounds in the original plant material used for marketing as well as in
392 different batches of marketed products and evaluate their possible contribution to the biological
393 activities of pharmaceutical interest.

394

395

CONCLUSIONS

396 It is known that plant species and, consequentially, the herbal preparations obtained from
397 them may vary considerably in the amounts of their chemical constituents such as specialized

398 (secondary) metabolites, which are mostly associated with their diverse biological activities and
399 consequent effects on human health. Besides comparing, for the first time, the total phenolic
400 content and flavonoid content of four different species of the genus *Petasites*, the results of our
401 study also give insights into the possible variations of these compounds in the two species used
402 in European folk medicine, *P. hybridus* and *P. albus*. The former species is especially pharma-
403 ceutically important as it is contained in herbal medicines and dietary supplements in the form
404 of standardized leaf or rhizome extracts. Our evaluation was done on extracts prepared from
405 leaves as a more ecologically sustainable source of *Petasites* bioactive compounds that, from a
406 health perspective, may also be more appropriate considering their initially lower pyrrolizidine
407 alkaloid content. Antioxidant activity of prepared extracts observed in this study was in high
408 correlation with their total phenolic content. Considering that the biological effects of polyphe-
409 nols and flavonoids, which are potentially beneficial to health, could probably be added to the
410 effects of the bioactive sesquiterpenes (petasins) to which the marketed *Petasites* products are
411 standardized, and considering their possible variabilities observed in this study, it could also be
412 interesting to investigate these compounds in more detail in future studies.

413

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