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3	Original research paper
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5	Phenolic content and antioxidant activity of Croatian and German honey
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ABSTRACT

Since honey has a therapeutic role in the treatment of many diseases, we investigated the 26 27 content of phenolic compounds and the antioxidant activity in acacia (Robinia pseudoacacia 28 L.), chestnut (Castanea sativa Mill.) and lime-tree (Tilia spp.) honey originated from Croatia 29 and Germany. In total phenols, flavonols and flavonols was observed higher values for Croatian 30 Castanea honey compared to German Castanea honey. Significant higher values of total 31 flavanols and hydroxycinnamic acids was measured in Croatian Tilia honey compared to 32 German *Tilia* honey. For *Robinia* honey, significant higher values of total phenols and flavonols 33 were observed in almost all Croatian honey samples compared to German honey. Croatian 34 honey samples had higher antioxidant activity compared to German honey samples with most 35 tested methods. The highest total phenols, total flavanols, ABTS, DPPH and FRAP values were measured in *Castanea* honey, then in *Robinia* honey and the lowest values in *Tilia* honey 36 37 samples. With new developed HPLC method, pinobanksin, pinocembrin and chrysin were 38 identified in the most honey samples. Our results imply that both botanical and geographical 39 origin influence the final quality of phenolic compounds and antioxidant activity in honey. High 40 positive correlation between the results of antioxidant activity and polyphenols were detected.

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Keywords: antioxidant activity, flavonoids, RP-HPLC, *Castanea sativa* honey, *Robinia pseudoacacia* honey, *Tilia* spp. honey

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INTRODUCTION

49 Honey is a natural, sweet and viscous mixture of substances created by honeybees, 50 processing nectar or honeydew (breaking down complex carbohydrates into simpler ones) in 51 their glands with saliva hydrolytic enzymes (1). Main components of honey are water, fructose 52 and glucose and then other sugars (maltose, sucrose and higher sugars), amino acids, proteins, 53 minerals, vitamins, organic acids and polyphenols (2). Flavonoids and phenolic acids are the 54 most common compounds from the polyphenol group and they are the most responsible for 55 honeys natural antioxidant activity and its protective effects on human health (3). In addition to 56 antioxidant properties, honey ingredients have anti-inflammatory, antimicrobial, 57 antimutagenic, antiparasitic and antitumor properties, and an increasing number of studies 58 indicate the importance of the therapeutic role of honey in the treatment of many diseases (4).

Nectar honey can be divided into unifloral or polyfloral honey, depending on whether the grazing of bees is directed mostly towards one or more plant species. More than 100 different types of unifloral honey are known in Europe (5). Persano Oddo *et al.* (5) analyzing the data from the International Honey Commission, reported 10 types of unifloral honey most commonly present in production and commercial availability in the European market.

The composition, quality and biological effects of honey depend on many production parameters. Some of them are plant origin of honey, geographical origin, species of bees that produce honey, climatic conditions as well as the technical process of honey processing, time age of the product and exposure to high temperatures. More precise control of these parameters affects the generation of high prices for honey production, which in turn leads to the problem of increased market share of counterfeits (6). 70 In order to eliminate counterfeits and ensure standards for honey products, important 71 properties of honey, the method of technological processing and methods for determining the 72 validity of the declaration and product quality are legally defined by Croatian Regulations (7) 73 and in the European Union by Harmonised methods of the European Commission (8, 9). 74 Melisopalinological analysis determines the composition of pollen in honey, and various 75 chemical analyzes define the composition of sugars, water, free acids, water-insoluble 76 substances, hydroxymethylfurfural content and electrical conductivity of honey and enzymatic 77 activity of diastase. In these analyzes, the analysis of the composition of phenolic compounds, 78 which are among the main carriers of antioxidant properties of honey, is not legally required 79 for testing the quality of honey. In the last ten years, there has been an increase in the number 80 of papers examining the composition of non-volatile components, such as polyphenols, which 81 significantly contribute to antioxidant activity. Some of these papers (1, 10–20) have researched 82 Croatian and other honeys from Europe investigating both total and individual polyphenol 83 content as well as antioxidant activity. According to Brščić et al. (21), consumers in Croatia 84 prefer unifloral acacia honey the most (56 %), followed by multifloral floral honey (44 %) and 85 meadow honey (35 %). They also prefer unifloral sage (25 %), chestnut (21 %) and linden (16 86 %) honey. A mild flavor (52 %) and brighter color (44 %) of honey are also preferred by 87 Croatians consumers. Based on these preferences, we decided to use two unifloral honeys in 88 our work: acacia and linden honey, which have mild flavors and brighter colors, and chestnut 89 honey, which has a stronger aroma and darker color. Additionally, these three types of honey 90 were available in both Croatia and Germany.

The aim of our study is to a) determine the polyphenol content and composition by HPLC analysis and by spectrophotometric determination of total soluble phenols (TP) by Foline-Ciocalteau reagent, total flavonoids (TF) by AlCl₃ method, total hydroxycinnamic acids (THA) and total flavonols (TFL) by HCl method and the total flavanol (TFLA) content by *p*-

95 dimethylaminocinnamaldehyde (DMACA) method and antioxidant activity (ABTS: 2,2'-azino-96 bis(3-ethylbenzothiazoline-6-sulfonic acid), DPPH: 2,2-diphenyl-1-picrylhydrazyl and FRAP: 97 ferric ion reducing antioxidant power) in acacia (Robinia pseudoacacia L.) honey, chestnut (Castanea sativa Mill.) honey and lime-tree (Tilia spp.) honey originated from Croatia and 98 99 Germany, b) compare the obtained results on the basis of plant botanical and geographical 100 origin. To determine the polyphenol content and composition, we have developed a new HPLC 101 method for detecting flavonoids in honey samples. The novelty of the research lies in the fact 102 that for the first time the same honey type produced in different geographical and climatic 103 regions was compared from Croatia and Germany. So far, the honey samples were compared from the point of botanical origin (11, 18, 19, 22–24), or from the point of production seasons 104 105 (10). In our study, we took into account both, geographical and botanical origin.

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- 107

EXPERIMENTAL

108 *Honey samples*

Seven samples of honey from the area of Central Europe were collected, four from Croatia and three from Germany. All samples were purchased from beekeepers, only certain samples were available as a commercial product in stores (Table I). All samples have the same method of technological processing, which was checked when buying honey and is legally defined in Europe with the Codex Alimentarius and the European Honey Directive. The origin of each honey is shown on a geographical map created with the help of the program QGIS 2.18 (Fig. 1). Samples were stored two weeks at room temperature in dark place to purification.

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117 Tabale I. Botanical and geographical origin of Croatian and German honey samples and their
118 commercial availability

-	Botanical origin	Geographical	Sample	Commercial
		origin	name	availability
-	Castanea sativa Mill.	Topusko, Croatia	CsTC	_
	Robinia pseudoacacia L.	Topusko, Croatia	RpTC	_
	Robinia pseudoacacia L.	Konjščina, Croatia	RpZC	+
	<i>Tilia</i> spp.	Bilogora, Croatia	TsBC	_
	Castanea sativa Mill.	Brandenburg,	CsBG	+
		Germany		
	Robinia pseudoacacia L.	Saska, Germany	RpSG	+
_	<i>Tilia</i> spp.	Saska, Germany	TsSG	+







123 Purification was performed as described in Kenjerić et al. (10) with slight modification. 124 Before sampling, each honey is well mixed. Then 25 g of each sample was dissolved in 125 mL 0.01 mol L^{-1} solution of hydrochloric acid. The solution is then vacuum filtrated via a 125 polyetersulphonic (PES) filter with pores of 0.2 µm. Chromatography with the aim of extracting 126 127 phenolic compounds from samples was performed in glass column (25×2 cm) with the help 128 of the stationary phase AMBERLITE[®] XAD[®] 2, SUPELCO[®]. About 50 g of stationary phase 129 per sample was washed off 15 minutes by 96 % ethanol (V/V), twice dH₂O and once with 150 130 mL 0.01 mol L^{-1} of hydrochloric acid solution. After passing the samples, the stationary phase with adsorbed compounds was washed with 250 mL 0.01 mol L^{-1} with a solution of 131 132 hydrochloric acid and 250 mL dH₂O. The elution of compounds from the stationary phase was 133 carried out with 175 mL of 96 % ethanol (V/V). The collected fractions are paired on a rotavapor 134 at a temperature range of 32–35 °C with a rotation of 60 rpm. The final masses obtained by 135 extraction were mixed with 96 % ethanol (V/V) so that the final mass concentration of each sample was 60 mg mL⁻¹. For further analysis, extracts were prepared at a mass concentration 136 of 10 mg mL⁻¹ and were purified three times by centrifugation with 5 min cycles on 15.000 g 137 138 and 4 °C. The prepared extracts are stored at a temperature of -20 °C until use.

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140 *Chemicals and apparatus*

141 Commercial polyphenol standards were purchased from Sigma-Aldrich GmbH (Germany) 142 and Extrasynthese (France). All chemicals and reagents were of analytical grade and supplied 143 by Sigma Aldrich GmbH (Germany) or Kemika (Croatia). RP-HPLC analyses were performed 144 using the Agilent 1100 Series system equipped with a quaternary pump, multiwave UV/Vis 145 detector, autosampler, fraction collector, analytical Zorbax Rx-C18 guard column (4.6 \times 12.5 146 mm, 5 µm particle size) and Poroshell 120 SB-C18 column (4.6 \times 75 mm, 2.7 µm particle size) 147 (Agilent Technologies, Waldbronn, Germany). All absorbance measurements of polyphenols were performed with a NanoDrop 2000c (Thermo Scientific[®]) and of antioxidant activity with
a Fluostar Optima microplate reader (BMG Labtech GmbH, Germany).

150

151 Spectrophotometric determination of polyphenols

Total soluble phenols (TP) of honey samples were determined with Foline-Ciocalteau reagent adapted for small volume as described in Vujčić Bok *et al.* (25). A volume 2 μ L of tested honey extracts was diluted with 158 μ L of distilled water and then 10 μ L of Foline-Ciocalteau reagent was added. Afterwards, 30 μ L Na₂CO₃ (1.88 mol L⁻¹) was added and the mixture was incubated for 30 min at 45 °C. The absorbance of the mixture was measured at 740 nm. The TP content was calculated from the calibration curve and expressed as gallic acid equivalents (GAE).

The content of total flavonoids (TF) of of honey extracts was determined with AlCl₃ adapted for small volume as described in Vujčić Bok *et al.* (25). To dilute the tested solution (2 μ L in 80 μ L of dH₂O), a volume of 6 μ L NaNO₂ (5 %) was added. After 5 min incubation, volume of 6 μ L AlCl₃ (10 %) was added and mixture was incubated at room temperature for additional 6 min. Afterwards, 40 μ L NaOH (1 mol L⁻¹) and distilled water were added to final volume of 200 μ L. The absorbance of the reaction mixture was read at 520 nm. The TF content was calculated from the calibration curve and expressed as quercetin equivalents (QE).

Total hydroxycinnamic acids (THA) and total flavonols (TFL) of honey extracts were measured as described in Vujčić Bok *et al.* (26) adapted for small volume using caffeic acid and quercetin as standards. Volume of 0.25 mL of the extract was mixed with 0.25 mL HCl (1 $g L^{-1}$; prepared in ethanol) and 4.55 mL HCl (2 g L⁻¹). The absorbance of the solution was read at 320 and 360 nm, respectively. THA and TFL contents were calculated from the 171 corresponding calibration curves and expressed as caffeic acid (CAE) and quercetin equivalents172 (QEE), respectively.

173 The total flavanol (TFLA) content was determined using *p*-dimethylaminocinnamaldehyde 174 (DMACA) adapted for small volume as described in Rusak *et al.* (27). A volume of 100 μ L of 175 tested honey extracts was mixed with 150 μ L of DMACA solution (0.1 % in 1 mol L⁻¹ HCl in 176 MeOH). After 10 min of incubation at room temperature, absorbance at 640 nm was measured. 177 TFL content was calculated from the calibration curve and expressed as catechin equivalents 178 (CE).

179

180 *RP-HPLC analysis of flavonoids*

181 Before HPLC analysis, honey samples were hydrolyzed as follows: 150 μ L of each extract 182 was mixed with 16,97 μ L of HCl (36.5 %, *V/V*) and incubated for 2 h at 80 °C and 300 rpm, 183 stored at -20 °C and centrifuged 15 min on 15.000 g until HPLC analysis.

184 Qualitative and quantitative RP-HPLC analyses of honey extracts were performed using the 185 Agilent 1100 Series system. The solvents used were: (A) 0.2 % (V/V) aqueous glacial acetic 186 acid, and (B) 80 % (V/V) methanol + 0.2 % (V/V) glacial acetic acid. Gradient profile was (A/B): 187 85/15 at 0 min, 51.7/48.3 at 20 min, 46.5/53.5 at 24 min, 36.5/63.5 at 30 min, 0/100 at 37.3 188 min, 0/100 at 40 min. 100/0 at 43 min. Injection volume was 15 µL, the constant flow rate 1.0 mL min⁻¹, and the column temperature was set at 30 °C. The multiwave UV/Vis detector was 189 190 set at 254, 280, 310, 335 and 360 nm. Phenolic compounds were characterized according to 191 their retention times and UV spectra compared with commercial standards. For the quantitative analyses, calibration curves were obtained by injection of 8 known concentrations (in the range 192 193 1-250 µg mL⁻¹) of the mixed 96 % EtOH standard solution in triplicate. The injection volume 194 was 15 µL. The honey extracts were compared with available phenolic standards (pinobanksin, 195 pinocembrin, chrysin, *p*-coumaric acid, syringic acid, chlorogenic acid and quercetin). The 196 results were expressed as μ g mL⁻¹ of honey weight.

197

198 Antioxidant activity

The ABTS (2,20-azinobis(3-ethylbenzothiazoline-6-sulfonic acid) assay was carried out as described in Radić Brkanac *et al.* (28). A volume of 2 µL of the tested honey extracts was added to 200 µL of ABTS solution and incubated for 6 min at room temperature. The absorbance of the reaction mixture was read at 740 nm. The radical scavenging activity was calculated as percentage of ABTS inhibition as follows: % inhibition = $[(A_0 - A_t)/A_0] \times 100$, where A_0 was the absorbance of the control (blank, without tested solution) and A_t was the absorbance in the presence of the tested solution.

206 DPPH (1,1-diphenyl-2-picrylhydrazyl) assay was performed as described in Radić Brkanac 207 *et al.* (28); 10 μ L of tested honey extracts was added to 190 μ L of freshly prepared ethanolic 208 DPPH solution (0.1 mmol L⁻¹) and incubated in the dark for 30 min at room temperature. The 209 decrease in absorbance was measured at 520 nm and the radical scavenging capacity was 210 calculated using the above-mentioned equation.

The ferric reducing antioxidant power (FRAP) assay was performed as described in Radić Brkanac *et al.* (28). The tested honey extracts (10 µL) was mixed with the 190 µL of freshly prepared FRAP reagent (and the absorbance was measured at 595 nm after 4 min of reaction time. The percent of Fe³⁺-TPTZ reduction was calculated using the formula: % reduction = [(A_t $-A_0$)/ A_t] × 100, where A_0 was the absorbance of the control (blank, without tested solution) and A_t was the absorbance in the presence of the tested solution. Trolox was used as a positive control for all antioxidant activity methods.

219 Statistical analysis

All results were evaluated using Statistica 13.3 software package (Stat Soft Inc., USA). RP-HPLC and results from spectrophotometric determination were subjected to one-way ANOVA for comparison of means and significant differences were calculated according to Duncan's multiple range test. The data are presented as the mean \pm standard deviations (SD). Pearson's correlation coefficient and Principal component analysis (PCA) between individual and total polyphenols and antioxidant activity were performed. Data were considered statistically significant at $p \le 0.05$.

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RESULTS AND DISCUSSION

- 229 Spectrophotometric determination of polyphenols
- 230 Total phenols (TP), total flavonoids (TF), total flavanols (TFLA), total flavonols (TFL) and
- total hydroxycinnamic acids (THA) in Croatian and German honey were presented in Fig. 2.



233 Fig. 2. Phenolic content: a) total phenolics (TP); b) total flavonoids (TF); c) total flavanols 234 (TFLA); d) total flavonols (TFL); e) total hydroxycinnamic acids (THA) in Croatian and 235 German honey. Values represent mean \pm SD of 3 replicates. Different letters indicate significant 236 difference at p < 0.05. Castanea sativa honey, Topusko, Croatia – CsTC; Castanea sativa 237 honey, Brandenburg, Germany - CsBG; Tilia spp. honey, Bilogora, Croatia - TsBC; Tilia spp. 238 honey, Saska, Germany - TsSG; Robinia pseudoacacia honey, Topusko, Croatia - RpTC; 239 Robinia pseudoacacia honey, Konjščina, Zagorje, Croatia - RpZC; Robinia pseudoacacia 240 honey, Saska, Germany – RpSG.

241

The highest TP (Fig. 2a) content was measured in *Castanea sativa* honey originated from Topusko, Croatia and the lowest in *Tilia* spp. honey originated from Bilogora, Croatia and Saska, Germany. Croatian *Castanea* honey had statistically higher values of TP than German *Castanea* honey. Same trend was observed between Craotian *Robinia* honey (Topusko and Konjščina, Zagorje) and German *Robinia* honey from Saska. No significant difference in TP between *Tilia* spp. honey from Bilogora, Croatia and Saska, Germany was detected.

248 In German Robinia honey from Saska were detected highest values of TF (Fig. 2b) and the 249 lowest in Tilia spp. honey originated from Bilogora (Croatia) and Saska (Germany). 250 Statistically significant decreased between all Robinia honey samples were observed as follows 251 RpSG, RpZC and then RpTC. All Robinia honey samples had higher values compared to all 252 *Castanea* and *Tilia* honey samples. *Castanea* honey samples had higher TF values compared to 253 *Tilia* honey samples. No significant difference between all *Tilia* spp. honeys was observed with 254 TF method. Also, no significant difference in TF between all Castanea sativa honeys was 255 observed.

Castanea sativa honey from Topusko (Croatia) had the highest TFLA (Fig. 2c) values. In *Tilia* spp. honey from Saska (Germany), TFLA was not detected. Significant higher values of
TFLA were observed between CsTC and CsBG and between RpZC and other *Robinia* honey
samples (RpTC and RpSG).

In *Robinia* honey from Saska (Germany) was detected highest values of TFL (Fig. 2d) and the lowest in all *Tilia* spp. honey samples (TsBC and TsSG). German *Robinia* honey (RpSG) had significant higher values than Croatian *Robinia* honey (RpTC and RpZC). Croatian *Castanea* honey (CsTC) had significant higher values than German *Castanea* honey (CsBG). No significant difference in TFL between all *Tilia* spp. honey (TsBC and TsSG) was detected.

Croatian *Tilia* honey (TsBC) had the highest THA (Fig. 2e). The lowest THA values were observed in all *Robinia* honey samples (RpTC, RpZC and RpSG). Significant decline in THA was measured as follows TsBC, and then in TsSG, then in CsTC and CsBG, and then in all *Robinia* honey samples (RpSG, RpZC and RpTC). In Croatian *Tilia* honey significant higher values of THA was observed compared to German *Tilia* honey. No significant difference for THA between all *Castanea* honey samples was observed. Also, no significant difference for THA between all *Robinia* honey samples was observed.

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273 Antioxidant activity

In Fig. 3, antioxidant activity (ABTS; % inhibition and mmol L^{-1} TE g^{-1} , DPPH; % inhibition and mmol L^{-1} TE g^{-1} and FRAP; % reduction Fe³⁺, mmol L^{-1} Fe²⁺ g^{-1} and mmol L^{-1} TE g^{-1}) of Croatian and German honey were presented.





Fig. 3. Antioxidant activity: a) ABTS; b) DPPH; c) FRAP in Croatian and German honey.
Values represent mean ± SD of 3 replicates. Different letters indicate significant difference at *p* < 0.05. Statistic is performed separate for results presented in % and separate for results
presented in mM. *Castanea sativa* honey, Topusko, Croatia – CsTC; *Castanea sativa* honey,
Brandenburg, Germany – CsBG; *Tilia* spp. honey, Bilogora, Croatia – TsBC; *Tilia* spp. honey,
Saska, Germany – TsSG; *Robinia pseudoacacia* honey, Topusko, Croatia – RpTC; *Robinia*

284 pseudoacacia honey, Konjščina, Zagorje, Croatia – RpZC; Robinia pseudoacacia honey,
285 Saska, Germany – RpSG.

286 The highest antioxidant activity was measured in Castanea honey (CsTC, CsBG) and Robinia honey (RpZC) with ABTS method (expressed as a percentage of inhibition and in 287 mmol L^{-1} TE g^{-1}) and the lowest in *Tilia* honey (TsBC). Sample RpZC had statistically higher 288 289 ABTS values then RpTC sample. Significant decrease in ABTS was observed in Robinia honey originated from Croatia (RpZC, RpTC) compared to German honey (RpSG). No significant 290 291 difference in antioxidant activity measured with ABTS (% inhibition and mmol L^{-1} TE g^{-1}) between all Castanea honey samples was observed. Also, no significant difference in 292 antioxidant activity measured with ABTS (% inhibition and mmol L^{-1} TE g^{-1}) between all *Tilia* 293 294 honey samples was observed.

In Croatian *Castanea* honey (CsTC) was measured the highest antioxidant activity with DPPH method (% inhibition and mmol L^{-1} TE g^{-1}) and lowest in German *Tilia* honey (TsSG). Significant decrease with DPPH (% inhibition and mmol L^{-1} TE g^{-1}) was observed in *Robinia* honey originated from Croatia (RpZC, RpTC) compared to German honey (RpSG). This trend was also observed between Croatian and German *Castanea* honey with DPPH method (% inhibition and mmol L^{-1} TE g^{-1}). Honey sample RpTC had statistically higher DPPH values then RpZC sample.

302 *Castanea* honey (CsTC) originated from Croatia had the highest antioxidant activity with 303 FRAP method (% reduction, mmol L^{-1} Fe²⁺ g⁻¹ and mmol L^{-1} TE g⁻¹) and lowest had German 304 *Tilia* honey (TsSG). All Croatian honey samples had statistically higher FRAP values (% 305 reduction, mmol L^{-1} Fe²⁺ g⁻¹ and mmol L^{-1} TE g⁻¹) then German honey samples.

306

307 RP-HPLC flavonoids

With new RP-HPLC method, we identified 3 flavonoids (Fig. 4, Table II) from 7 available phenolic standards (flavonoids: pinobanksin, pinocembrin, chrysin and quercetin; phenolic acids: *p*-coumaric acid, syringic acid and chlorogenic acid). The novelty of the HPLC method refers to a new solvent gradient adapted to the honey. The solvent gradient is described in Experimental. New method is shorter (43 min compared to 60 min in the study of Kenjerić *et al.*(10) and Šarić *et al.* (19); 52 min in the study of Tomás-Barberán *et al.* (22) and still separates both flavonoids and phenolic acids.







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320 Tabale II. Content of individual and total identified flavonoids by HPLC

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Flavonoid (µg mL⁻¹)

		Chrysin	Pinocembrin	Pinobanksin	TiF		
	Castanea sativa honey, Topusko, Croatia – CsTC Castanea sativa honey,	9.05 ^b	4.24 ^e	13.14 ^c	26.43 ^d		
	Brandenburg, Germany - CsBG	8.25 ^c	8.42 ^c	21.04 ^a	37.71 ^a		
	Bilogora, Croatia – TsBC Tilia spp. honey. Saska	6.92 ^e	8.27 ^c	21.43 ^a	36.61 ^b		
	Germany – TsSG <i>Robinia pseudoacacia</i>	4.74 ^f	7.89 ^d	16.51 ^b	29.13 ^c		
	honey, Topusko, Croatia – RpTC <i>Robinia pseudoacacia</i>	7.22 ^d	11.70 ^b	n.d.	18.91 ^e		
	honey, Zagorje, Croatia – RpZC <i>Robinia pseudoacacia</i>	17.04 ^a	19.53 ^a	n.d.	36.57 ^b		
	honey, Saska, Germany – RpSG	7.20 ^d	3.81 ^f	n.d.	11.01 ^f		
325 326	Flavonoids chrysin and pinot	cembrin were	identified in all t	ested Croatian and	d German		
327	honey samples. Flavonoid pinoba	nksin is detect	ted only in Croatian	n and German Cas	<i>tanea</i> and		
328	<i>Tilia</i> honey samples. The highest	chrysin and p	binocembrin conter	it was measured in	n Croatian		
329	Robinia honey sample (RpZC)	and lowest in	n <i>Tilia</i> honey san	ple originated from	om Saska		
330	(Germany) for chrysin and in Robi	<i>nia</i> honey orig	ginated from Saska	(Germany) for pin	ocembrin.		
331	Tilia spp. honey originated from Bilogora (Croatia) and Castanea sativa honey originated from						
332	Brandenburg (Germany) had the highest pinobanksin content. Total identified flavonids (TiF)						
333	were calculated as sum of identifi	ed chrysin, pi	nocembrin and pin	obanxin. Sample (CsBG had		
334	the highest values of TiF and in R	pSG were det	ected lowest value	s of TiF.			
335							

For TP and almost all antioxidant methods (ABTS, FRAP and FRAP) expressed as a percentage of inhibition and in mmol L^{-1} , Croatian *Castanea* honey (CsTC) had the highest

Polyphenols and antioxidant activity

339 values, followed by German Castanea honey (CsBG) and Croatian Robinia honey (RpZC), 340 followed by Croatian Robinia honey (RpTC), followed by German Robinia honey (RpSG), 341 followed by Croatian Tilia honey (TsBC) and German Tilia honey (TsSG). Same trend was 342 observed for *Castanea* and *Robinia* honey and for *Castanea* and *Tilia* honey, and opposite for 343 Robinia and Tilia honey in literature (11, 24, 29–31). Based on sensory properties - given the 344 dark color, intense smell and taste, it was expected that the Castanea honey will have high 345 values of phenolic compounds and antioxidant activity. Gorjanović et al. (23) reported very 346 strong positive correlation between honey color, TP and antioxidant activity (FRAP, ORAC, 347 TEAC and DPPH). The quantitatively high values of the almost all results of the Castanea 348 honey can most probably be explained by the fact that chestnut honey is categorized into types 349 of honey with extremely high pollen content (85 %) because C. sativa Mill. honey plant is 350 characterized by hyperproduction of pollen and nectar (32) and pollen contributes to the content 351 of proteins, phenolic compounds, vitamins and minerals in honey and its antioxidant capacity 352 (33-35). Also, composition of collected nectar may influence on polyphenols and antioxidant 353 capacity of honey (20). The samples with the lowest antioxidant capacity, the content of total 354 phenols and the examined phenolic subgroups are German and Croatian *Tilia* honey samples. 355 The most probable reason for such results is the fact that *Tilia* honey according to Louveaux et 356 al. (32) is grouped into a type of uniforal honey with low pollen content (20-30 %). The 357 obtained results are in accordance with the assumption that *Tilia* honey due to its mild 358 organoleptic properties and yellowish transparent color will have a lower concentration of 359 phenolic compounds and antioxidant capacity compared to Castanea honey. The permitted 360 levels of pollen grains in *Tilia* honey can be 10 % if it possesses all the important organoleptic 361 properties. Robinia honey is also grouped into a type of unifloral honey with low pollen content 362 (20–30 %) (32). This is in accordance with the lower measured values for most of the tested 363 methods for *Robinia* honey compared to *Castanea* honey. According to the available literature,

there are no results of the spectrophotometric determination of total flavonoids (TF), total flavanols (TFLA), total flavonols (TFL) and total hydroxycinnamic acids (THA) in Croatian and German honey. These methods are rapid and low cost and, in the future, it would be desirable to apply them to determine the composition of polyphenolic groups of compounds because they allow testing purified and unpurified honey samples.

369 In Croatian Robinia honeys, Kenjerić et al. (10) detected six flavonoids (quercetin, luteolin, 370 kaempferol, apigenin, chrysin and galangin), and presence of phenolic acids (caffeic acid and 371 *p*-coumaric acid) was also confirmed. Flavonoids myricetin, quercetin, luteolin, kaempferol, 372 apigenin, isorhamnetin, chrysin and galangin were identified in Croatian Castanea honey 373 samples by Kenjerić et al. (36). According to the available literature, there are no results of the 374 Croatan Tilia honey flavonoid profile. Tomás-Barberán et al. (22) detected caffeic acid, p-375 coumaric acid, ferulic acid, quercetin, luteolin, kaempferol, pinobanksin, pinocembrin, and 376 chrysin in some German Robinia honey samples, caffeic acid, p-coumaric acid, pinobanksin, 377 pinocembrin and chrysin were detected in some German Castanea honey samples and p-378 coumaric acid, 8-methoxykaempferol and chrysin in German Tilia honey sample. In our study 379 we identified chrysin and pinocembrin in all tested Croatian and German honey samples and 380 pinobanksin only in Croatian and German Castanea and Tilia honey samples. Variability of 381 flavonoid profile and concentrations is to be expected due to the seasons, climatically conditions 382 and other factors.

383 The values of individual flavonoids obtained by HPLC analysis do not necessarily follow 384 the relationships obtained by measuring total phenols and antioxidant activity. The identified 385 chrysin, pinobanksin and pinocembrin are just some of the compounds that contribute to the 386 total phenol composition and antioxidant activity, so reported values of the mentioned 387 flavonoids give a more specific view of the mutual differences between honey samples. Thus, 388 for example, the sample with the highest concentration of pinobanksin is the Croatian *Tilia* honey and THA, although antioxidant activity and TP, TF, TFLA and TFL were the lowest
compared to other samples. Also, some of flavonoids such as Pcb do not have pronounced
antioxidant properties (37).

392 Based on the detected and identified flavonoids, we can assume the positive biological 393 effects of certain honey samples on human health. Croatian Robinia honey originated from 394 Zagorje had a high chrysin content compared to other honey samples which is why it could 395 have a positive effect in anti-inflammatory processes because chrysin inhibits cyclooxygenase-396 2, the enzyme responsible for inflammation and accompanying pain (38). Same sample had 397 highest pinocembrin content compared to other honey samples, which suggests its potentially 398 beneficial effect on cell protection due to poor blood circulation. According to Khalil et al. (37) 399 pinocembrin inhibits the onset of apoptosis in such cells. Croatian Tilia honey and German 400 *Castanea* honey had the highest pinobaksin content compared to other samples, which is why 401 it could have a positive effect against tumor formation. According to Silva-Carvalho et al. (39), 402 pinobanksin acts by slowing the growth of tumor cells.

403 In our study Castanea honey (CsTC, CsBG) and Robinia honey (RpZC and RpTC) had 404 moderate (42.07 %, 39.38 %, 33.99 % and 39.95 %), and all other sample weak (17.73–22.72 %) antioxidant activity in relation to Trolox (82.43 %) by ABTS method. Moderate (40.92-405 406 48.56 %) antioxidant activity with DPPH method was observed in Castanea honey (CsTC, 407 CsBG) and *Robinia* honey (RpTC; 28.75 %), and all other sample had weak (14.37–21.69 %) 408 antioxidant activity in relation to Trolox (82.06 %). With FRAP method, all tested Croatian and 409 German honey samples showed strong (73.33–94.29 %) antioxidant activity in relation to 410 Trolox (97.45 %). Our classification of antioxidant activity of honey samples is based on the 411 Vujčić et al. (40) classification. In this paper antioxidant activity is classified on weak (< 35 412 %), moderate (35–70 %) and strong (70–100 %) in relation to the positive control (100 %) for 413 herbal originated extracts.

414 Analyzed Croatian honey samples had higher level of polyphenols and stronger antioxidant 415 activity in comparison to German honey samples. Possible explanation lays in the fact that there 416 is greater diversity of Croatian flora and climatic characteristics are more favorable for 417 beekeeping in Croatia than in Germany. This supports the influence of geographical origin on 418 the quality of honey. If we observe climatic characteristics as the only factor influencing the 419 quality of honey in a geographical area, we can spot that with increase of northern altitude, 420 probably due to the decrease of average annual temperatures, the quality of honey also 421 decreases. This is explained by the lower activity of bees in collecting pollen and nectar at lower 422 temperatures because bees, instead of collecting and producing honey, spend most of their time 423 heating the hive to the optimum temperature of 33–35 °C (41). According to Köppen-Geiger 424 climate classification, Germany is characterized by moderately warm humid climate with warm 425 summers (Cfb), and Croatia by a moderately warm climate with hot summers without drought 426 (Cfa) and with dry summers (Csa) (41). For the period from 1901 to 2000, the Croatian average 427 temperature was 10.90 °C, and the German 8.50 °C (43). It is obvious from the above that 428 Croatia has higher average temperatures compared to Germany, which is logical if we take into 429 account its latitude and the influence of the Mediterranean Sea; the climate certainly remains 430 one of the factors contributing to the difference between Croatian and German honey. An 431 additional argument that is closely related to climatic characteristics is the trend of decreasing 432 diversity of flora from the equator to the north, which affects the quality of honey (44). By 433 reducing the biodiversity of flora in the range of bees, the availability of diverse pollen is 434 reduced, which causes a weak colony due to loss of nutrition and immunodeficiency caused by 435 non-diverse diet of bees (44). According to a direct comparison of flora according to data from 436 2001, Croatia has 5347 different types of vascular flora, while Germany has 2742 (46). Therefore, the flora of Croatia has 0.07561 species per km² of its area, while Germany has 437

438 0.00771 species per km², which means that on one km² within the Croatian territory bees will
439 theoretically have 89.80 % more varied pollen.

440

441 *Statistics*

- 442 Pearson's correlation coefficient between polyphenolic content and antioxidant activity of
- 443 Croatian and German honey is presented in Table III.

	445 Gern	nan honey											
TP	TF	TFLA	TFL	THA	Chr	Pcb	Pbs	TiF	ABTS	ABTS	DPPH	DPPH	FRAP
									%	mmol L ⁻¹ TE	%	mmol L ⁻¹ TE	%
1.00													
0.28	1.00												
0.78	-0.09	1.00											
0.43	0.84	0.33	1.00										
-0.49	-0.87	0.04	-0.66	1.00									
0.47	0.47	0.15	0.22	-0.46	1.00								
0.09	0.19	-0.34	-0.26	-0.32	0.76	1.00							
-0.23	-0.87	0.17	-0.70	0.85	-0.44	-0.37	1.00						
0.01	-0.58	0.04	-0.74	0.49	0.36	0.46	0.63	1.00					
0.94	0.26	0.59	0.24	-0.55	0.60	0.33	-0.20	0.21	1.00				
0.94	0.26	0.59	0.24	-0.55	0.60	0.33	-0.20	0.21	1.00	1.00			
0.89	-0.03	0.80	0.22	-0.22	0.12	-0.22	0.17	0.10	0.82	0.82	1.00		
0.89	-0.03	0.79	0.21	-0.23	0.14	-0.19	0.18	0.13	0.83	0.83	1.00	1.00	
0.92	0.47	0.58	0.50	-0.60	0.60	0.20	-0.31	0.04	0.89	0.89	0.78	0.79	1.00
0.98	0.27	0.80	0.42	-0.43	0.55	0.11	-0.16	0.12	0.93	0.93	0.87	0.88	0.94
0.98	0.27	0.80	0.42	-0.43	0.55	0.11	-0.16	0.12	0.93	0.93	0.87	0.88	0.94
ns are significant at $p < 0.05$													

444 Tabale III. Pearson's correlation coefficient between total and individual polyphenolic content and antioxidant activity of Croatian and 445 German honey

447 TP correlated very strongly (r > 0.80) with all antioxidant activity methods (ABTS %: 0.94, ABTS mmol L⁻¹ TE 0.94, DPPH % 0.89, DPPH mmol L⁻¹ TE: 0.89, FRAP % 0.92, FRAP 448 mmol L^{-1} Fe²⁺: 0.98 and FRAP mmol L^{-1} TE: 0.98) and strongly (r > 0.60 < 0.79) with TFLA 449 450 (0.78). Positive very strong correlation was observed between TF and TFL (0.84) and negative 451 very strong correlation was observed between TF and THA (-0.87) and TF and Pbs (-0.87). TFLA had very strong (0.80) correlation with DPPH %. FRAP mmol L^{-1} Fe²⁺ and FRAP mmol 452 $L^{-1}TE$ and strong (0.79) with DPPH expressed as mmol $L^{-1}TE$. THA correlated very strongly 453 454 with Pbs (0.85) and Chr correlated strongly with Pcb (0.76). All antioxidant methods had very strong (0.82–1) or strong correlation (DPPH % and FRAP %: 0.78, DPPH mmol L^{-1} TE and 455 456 FRAP %: 0.79) among themselves.

457 The first (Factor 1) and the second (Factor 2) principal components (PC) described 54.68
458 % and 23.94 % of the variance (Fig. 5).





461 Fig. 5. Principal component analysis of the measured polyphenols and antioxidant activity
462 in the Croatian and German honey. a) Score plot separating the Croatian and German Castanea,
463 Tilia and Robinia honey samples; b) the loading plot of polyphenols and antioxidant activity as

variables. *Castanea sativa* honey, Topusko, Croatia = CsTC; *Castanea sativa* honey,
Brandenburg, Germany – CsBG; *Tilia* spp. honey, Bilogora, Croatia – TsBC; *Tilia* spp. honey,
Saska, Germany – TsSG; *Robinia pseudoacacia* honey, Topusko, Croatia – RpTC; *Robinia pseudoacacia* honey, Konjščina, Zagorje, Croatia – RpZC; *Robinia pseudoacacia* honey,
Saska, Germany – RpSG and TiF – total identified flavonids, Pbs – Pinobanksin, Pcb –
pinocembrin, Chr – chrysin, total flavanols – TFLA, total flavonoids – TF, total flavonols –
TFL, total hydroxycinnamic acids – THA, total phenols – TP.

471

472 Together, the first two PCs represent 78.62 % of the total variability. With PCA plot (Fig. 473 5a) honey samples were divided on three groups of honey based on their botanical origin. So, the highest distance was detected between Castanea, Tilia and Robinia honey and the smallest 474 475 distance was detected between Croatian and German Tilia honey (TsBC and TsSG), then in 476 Croatian and German Castanea honey (CsTC and CsBG) and then between Croatian (RpZc 477 and RpTC) and German (RpSG) Robinia honey. Both, Croatian and German Castanea honey 478 (CsTC and CsBG) had strong loadings with most tested total (TP, TFLA and TIF) compounds and antioxidant activity (ABTS: % and mmol L^{-1} TE, FRAP: %, mmol L^{-1} TE and mmol L^{-1} 479 Fe^{2+} , DPPH: % and mmol L⁻¹ TE) (Fig. 5b). Croatian (RpZc and RpTC) *Robinia* honey had 480 481 strong loadings with total (TF and TFL) and individual (Chr and Pcb) polyphenolic compounds 482 (Fig. 5b). Strong loadings with Pbs and THA were detected in Croatian and German Tilia honey 483 (TsBC and TsSG) (Fig. 5b).

484 Correlation analysis confirmed the expected positive correlation between the results of 485 antioxidant methods with the total phenol content (TP) and phenolic subgroup (TFLA). 486 According to Moniruzzaman *et al.* (47) and Flanjak *et al.* (24), phenolic compounds are 487 responsible for antioxidant properties of honey. TF method showed low values of positive and 488 negative correlation coefficients, which are without statistical significance. This can be explained by the non-specificity of this method (48). The THA method shows a negative
correlation, *i.e.* an inversely proportional relationship with the results of antioxidant methods.
Since hydroxycinnamic acids are powerful antioxidants that can mediate scavenging of harmful
reactive oxygen species (49), our correlation results could be the result of certain non-specific
reactions.

494 Each honey declared as a unifloral type may contain a different percentage of pollen grains 495 due to the impossibility of direct control of bee grazing, so it is important to define the validity 496 of the declared botanical origin by melisopalinological analysis and by analyzing and defining 497 different markers, *i.e.* specific reference values of certain compounds in honey (50). Overview 498 of the similarities and differences between different Croatian and German Castanea, Tilia and 499 Robinia honey samples as well as the interrelationships between the measured properties 500 (polyphenol composition and antioxidant activity) were provided by the PCA plots for the 501 purpose of indirectly determining the botanical origin of honey. Three groups of honey 502 (Castanea, Tilia and Robinia) based on their botanical origin were divided with PCA. 503 According to results of PCA, we can conclude that *Castanea*, is the best quality unifloral honey 504 compared to Robinia and Tilia honey. Because, Castanea honey (CsTC and CsBG) had strong 505 loadings with most tested total (TP, TFLA and TIF) compounds and antioxidant activity (ABTS: % and mmol L^{-1} TE, FRAP: %, mmol L^{-1} TE and mmol L^{-1} Fe²⁺, DPPH: % and mmol 506 507 L^{-1} TE). From PCA plots, it can be seen that the different position of *Tilia* honey based on 508 ordinate is most affected by the content of THA and Pbs content, which for these honey samples 509 are the largest compared to other samples, while the position of Robinia honey is most affected 510 by TF and TFL and the content of Pcb and Chr. In almost all conducted analyzes, Croatian 511 unifloral types of honey are of better quality than German ones. In PCA plots (Fig. 5a,b), 512 Croatian honey samples are always grouped closely to the most of measured methods. Croatian 513 CsTC samples had smaller distance in PCA plots with TP, TFLA and antioxidant activity methods (ABTS: % and mmol L^{-1} TE, FRAP: %, mmol L^{-1} TE and mmol L^{-1} Fe²⁺, DPPH: % and mmol L^{-1} TE) compared to German CsBG samples. German CsBG samples had smaller distance in TiF compared to Croatian CsTC samples. *Robinia* honey samples originated from Croatia had smaller distance in PCA plots with TF, TFL, Chr, Pcb while German *Robinia* honey was from the opposite axis of the aforementioned methods. *Tilia* honey originated from Croatia had smaller distance in PCA plots with Pbs and THA compared to *Tilia* honey originated from Germany.

- 521
- 522

CONCLUSIONS

523 Through most methods, the sample with the highest antioxidant capacity and the 524 quantitative content of total phenolic compounds is Croatian chestnut honey. According to plant 525 origin, chestnut honey is of higher quality in terms of the quantitative composition of phenolic 526 compounds and antioxidant capacity than acacia and linden honey, and acacia honey is of better 527 quality than linden honey. The above order of unifloral types of honey is most likely caused by 528 the amount of pollen present in the honey because pollen contributes to the final content of 529 phenolic compounds, and thus to the antioxidant capacity of the honey. Additionally, the 530 composition of the collected nectar may influence polyphenols and antioxidant capacity of 531 honey. The analyzed Croatian honey samples are of better quality in terms of the composition 532 of phenolic compounds and antioxidant capacity compared to the German honey samples, most 533 likely due to more favorable climatic characteristics for bee breeding and greater diversity of 534 Croatian flora.

535

Acronyms, abbreviations, symbols. – ABTS – 2,2 -azinobis(3- ethylbenzothiazoline-6 sulfonic acid), CsBG – Castanea sativa honey, Brandenburg, Germany, CsTC – Castanea
 sativa honey, Topusko, Croatia, FRAP – Ferric Reducing/Antioxidant Power Assay, RpZC –

539	Robinia pseudoacacia honey, Konjščina, Zagorje, Croatia, RpSG - Robinia pseudoacacia
540	honey, Saska, Germany, RpTC – Robinia pseudoacacia honey, Topusko, Croatia, TiF – total
541	identified flavonids, TsBC - Tilia spp. honey, Bilogora, Croatia, TsSG - Tilia spp. honey,
542	Saska, Germany, TFLA – total flavanols, TF – total flavonoids, TFL – total flavonols, THA –
543	total hydroxycinnamic acids, TP – total phenols.
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