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Exploration of the chemical space of benzamide-based voltage-gated potassium channel Kv1.3 inhibitors

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ABSTRACT

The voltage-gated potassium channel $K_v 1.3$ is a key regulator of T-cell activation and a validated therapeutic target for autoimmune and inflammatory diseases. In this study, a ligand-based design strategy was employed to expand a library of benzamide-derived $K_v 1.3$ inhibitors. Starting from a previously optimized thiophene-based inhibitor, structural modifications were introduced to the 2-methoxybenzamide moiety and the central tetrahydropyran or cyclohexane scaffold. A series of ketone, hydroxy, and carbamate derivatives was synthesized and evaluated for $K_v 1.3$ inhibition using whole-cell patch-clamp electrophysiology. Structure–activity relationship analysis revealed that *cis*-isomers in the hydroxy series exhibited stronger activity

than their *trans* counterparts, with some analogs displaying submicromolar IC_{50} values. In the carbamate series, *trans*-isomers were generally more potent, with *trans*-18 and *trans*-16 achieving IC_{50} values of 122 nM and 166 nM, respectively. These results provide valuable insights into the design of K_v1.3 inhibitors and support further development of these compounds for immunomodulatory applications.

Keywords: Kv1.3, inhibitor, ion channel, ligand-based design, molecular modeling

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INTRODUCTION

Potassium channels are transmembrane proteins that facilitate the selective passage of potassium ions across the plasma membrane following their electrochemical gradient. These channels are classified into different families, with voltage-gated potassium channels (K_V) representing a major group. The $K_V 1.x$ (Shaker) subfamily, the largest within this group, consists of eight members ($K_V 1.1-K_V 1.8$) (1, 2). $K_V 1.3$ is widely expressed throughout the human body and plays a crucial role in numerous cellular processes. It is located in the plasma membrane, the inner mitochondrial membrane (mito $K_V 1.3$), the nuclear membrane, and the membrane of the *cis*-Golgi apparatus (3). $K_V 1.3$ expression has been identified in multiple cell types including neurons, osteoclasts, epithelial cells, and immune cells such T- and B-lymphocytes, macrophages and microglia, where it contributes to the regulation of membrane potential and calcium signaling (2).

Given the well-established role of T-cells in the pathogenesis of autoimmune diseases, $K_v 1.3$ has emerged as an attractive therapeutic target. It is highly expressed in effector memory T (T_{EM}) cells, which mediate autoimmune disorders, as well as in other leukocytes involved in chronic inflammatory diseases, such as B-lymphocytes, macrophages, microglia, and dendritic cells. Inhibition of $K_v 1.3$ has been shown to induce membrane depolarization, thereby preventing T-cell proliferation and cytokine production. Consequently, the development of selective $K_v 1.3$ inhibitors that specifically target disease-inducing T_{EM} cells while preserving normal immune function could be beneficial in treating autoimmune and chronic inflammatory conditions, including psoriasis, multiple sclerosis, type 1 diabetes mellitus, atherosclerosis, asthma, and rheumatoid arthritis (4, 5).

 $K_V 1.3$ has also been identified as a promising molecular target for anticancer therapy due to its involvement in critical cellular processes, including proliferation, calcium signaling, cell volume regulation, adhesion, migration, apoptosis, and invasion (6). Although $K_V 1.3$ is increasingly recognized as a tumor marker, a definitive pattern distinguishing its expression in cancerous versus healthy cells has not yet been established, as its levels appear to be influenced by the tumor type and disease stage (7). Nevertheless, aberrant $K_V 1.3$ expression has been observed in breast, colon, and prostate tumors, as well as in smooth muscle and skeletal muscle cancers. It is also present in mature neoplastic B-cells in chronic lymphocytic leukemia, with a notable correlation between its expression and mitochondrial localization.⁶ Potential therapeutic strategies for cancer treatment involve selectively inhibiting cancer cell proliferation or inducing apoptosis (8). Induction of apoptosis in cancer cells by $K_V 1.3$ inhibition might be considered as an effective method to selectively kill cancer cells (9). Beyond its roles in immunity and oncology, $K_V 1.3$ has been implicated in pathways regulating energy homeostasis and body weight, making it a potential target for obesity treatment (10).

One of the primary challenges in developing Kv1.3 inhibitors lies in the high sequence homology among Kv1.x family members, making it difficult to achieve potent and selective inhibition of Kv1.3. Several Kv1.3 inhibitors have been designed to specifically target Kv1.3 in the plasma membrane. Among them, the psoralen derivative PAP-1 (**1**, Fig. 1) is currently the most potent (IC₅₀ = 2 nM) and selective (*i.e.* 23-fold over Kv1.5) small-molecule Kv1.3 inhibitor (11). The antimycobacterial drug clofazimine (**2**, Fig. 1) is another well-characterized Kv1.3 inhibitor (IC₅₀ = 300 nM), displaying tenfold selectivity over Kv1.1, Kv1.2, Kv1.5, and Kv3.1 (12). Another important Kv1.3 inhibitor, a benzamide derivative known as PAC (**3**, Fig. 1), was identified through a high-throughput screening campaign. It exhibited an IC₅₀ of 200 nM but lacked selectivity among Kv1.x family members. *In vitro* functional assays demonstrated that this compound reversibly inhibits calcium-dependent T-cell activation and suppresses IL-2 production in a concentration-dependent manner without inducing cytotoxicity or affecting calcium-independent T-cell stimulation pathways (13).



Fig. 1. Structures of known representative $K_V 1.3$ inhibitors PAP-1 (1), clofazimine (2) and PAC (3).

Since the binding site of benzamide-based $K_v 1.3$ inhibitors remains unidentified, we employed a ligand-based drug design strategy, utilizing a 3D similarity search of previously reported benzamide inhibitors. This approach led to the discovery of a novel thiophene-based compound **A**, which demonstrated selectivity for $K_v 1.3$ channels (14). Further structural modifications of compound **A** resulted in the development of the 3-thiophene-based inhibitor **B** (Fig. 2), which demonstrated an IC₅₀ of 470 nM and an 18-fold selectivity over related $K_v 1.x$ family channels in *Xenopus laevis* oocytes (15). Additional optimization led to compound **C**, which inhibited $K_v 1.3$ -mediated currents in activated human T-lymphocytes with an IC₅₀ value of 26.1 nM (16). In this work, we further investigate the structure-activity relationship (SAR) of this series by extensively modifying the 2-methoxybenzamide moiety of the inhibitor while retaining the 3-thienyl group at position 4 of the core tetrahydropyran or cyclohexane ring (Fig. 2).



Fig. 2. Optimization of $K_v 1.3$ inhibitor **A**, identified by virtual screening, to more potent analogs **B** and **C**, and further structure-activity relationship investigation in this work.

EXPERIMENTAL

Chemistry – General

The reagents and solvents used were obtained from commercial sources (i.e., Acros Organics, Sigma-Aldrich, TCI Europe, Merck, Carlo Erba, Apollo Scientific) and were used as provided. Analytical thin-layer chromatography (TLC) was performed on silica gel aluminum sheets (60 F₂₅₄, 0.20 mm; Merck). Flash column chromatography was performed on silica gel 60 (particle size 0.040–0.063 mm, Merck). ¹H NMR and ¹³C spectra were recorded at 400 and 100 MHz, respectively, on a Bruker Avance III NMR spectrometer (Bruker, MA, USA) at 295 K. The chemical shifts (δ) are reported in ppm and are referenced to the deuterated solvent used. HRMS measurements were performed on a LC-MS/MS system (Q Executive Plus; Thermo Scientific, MA, USA). Mass spectrometry measurements were performed on an Expression CMS^L mass spectrometer (Advion, USA). Analytical reversed-phase UPLC analyses were performed using a modular system (Thermo Scientific Dionex UltiMate 3000 modular system; Thermo Fisher Scientific Inc., USA). Method: Waters Acquity UPLC® HSS C18 SB column $(2.1 \times 50 \text{ mm}, 1.8 \mu\text{m}), t = 40 \text{ }^{\circ}\text{C};$ injection volume = 5 $\mu\text{L};$ flow rate = 0.4 mL min⁻¹; detector $\lambda = 254$ nm; mobile phase A (0.1 % trifluoroacetic acid (TFA) [v/v] in water), mobile phase B acetonitrile (MeCN). Gradient: 0-2 min, 10 % B; 2-10 min, 10 %-90 % B; 10-12 min, 90 % B. Purities of the tested compounds were established to be ≥ 95 % at 254 nm, as determined by UPLC. The syntheses of the compounds are illustrated in Schemes 1 and 2, experimental procedures and characterization data are provided in the Supporting information.



Scheme 1. Synthesis of the target compounds **11a** and **11b**.



Scheme 2. Synthesis of the target compounds 13a-i, *trans*-14a-d, *trans*-14g, *trans*-14i, *cis*-14a-d, *cis*-14g, *cis*-14i, *trans*-15-20 and *cis*-21.

Patch-clamp electrophysiology

Mouse L929 fibroblasts stably expression $mK_v 1.3$, were a gift from Dr. K. George Chandy (University of California, Irvine, USA). All experiments were conducted with an EPC-10 amplifier (HEKA, Lambrecht/Pfalz, Germany) in the whole-cell configuration with a holding potential of -80 mV. Pipette resistances averaged around 2.5 MΩ. Compound solutions were prepared fresh in Na⁺ Ringer from 10 mM stock solutions in DMSO directly before the experiments. For current measurements, we used an internal pipette solution containing 160 mM KF, 2 mM MgCl₂, 10 mM HEPES, and 10 mM EGTA, with a pH of 7.2 and an osmolarity of ~ 300 mOsm. Sodium Ringer was used as an external solution containing the following: 160 mM NaCl, 4.5 mM KCl, 2 mM CaCl₂, 1 mM MgCl₂, and 10 mM HEPES, with a pH of 7.4 and an osmolarity of ~ 300 mOsm. Currents were elicited with a 200-ms voltage step to +40 mV, followed by 45 seconds of holding at a resting membrane potential of -80 mV. A usedependence protocol, whereby cells were pulsed to 40 mV every 1 second, was used prior to the step protocol for compound to ensure that channel kinetics were as expected. If currents exceeded 2 nA 60-80 % series resistance compensation was used. Concentration dependent current inhibition as measured as reduction of area under the current curve was fitted with the Hill equation using GraphPad Prism 8 (GraphPad Software, USA). All data points represent at least 3 independent experiments and are presented as mean \pm standard deviation (SD). IC₅₀ values are reported with 95 % confidence intervals (CI).

RESULT AND DISCUSSION

Chemistry

Based on our previously published compounds (15, 16), we further investigated the SAR for K_v1.3 inhibition (Fig. 2). We synthesized compounds **11a** and **11b**, in which the 3-thiophene and tetrahydropyran moieties were retained, while the 2-methoxyphenyl group of compound **B** was replaced with either a 2-methoxycyclohexyl or 3-methoxythiophenyl moiety, respectively. Next, the tetrahydropyran ring was substituted with a cyclohexane ring bearing keto, hydroxy, or carbamate groups. Compounds **13a–i** contain a ketone at the 4-position of the cyclohexane ring and feature various modifications of the 2-methoxybenzamide moiety. Reduction of the ketone in compounds **13a–h** produced the corresponding hydroxy analogs: *trans*-**14a–d**, *cis*-**14a–d**, *trans*-**14g**, *cis*-**14g**, *trans*-**14i**, and *cis*-**14i**. These hydroxy compounds were then further

modified to generate methylcarbamate-, propylcarbamate-, and 3-methoxypropylcarbamatecontaining derivatives **15–21**. All hydroxy- and carbamate-substituted compounds were obtained as diastereomeric mixtures, which were subsequently separated into their *cis*- and *trans*-isomers using column chromatography. The synthetic routes for these new compounds are outlined in Schemes 1 and 2.

In Scheme 1, thiophene-3-acetonitrile was reacted with 1-chloro-2-(2-chloroethoxy)ethane in the presence of sodium hydride (NaH, 60 % dispersion in mineral oil) under an argon atmosphere at room temperature, yielding the intermediate 4-(thiophen-3-yl)tetrahydro-2*H*-pyran-4-carbonitrile (**4**). The nitrile group was subsequently reduced using lithium aluminum hydride (LiAlH₄) in tetrahydrofuran (THF) to produce the primary amine **9**. This amine was then reacted with either 4-methoxythiophene-3-carbonyl chloride or 2-methoxycyclohexane-1-carbonyl chloride to afford the final compounds **11a** and **11b**.

For the synthesis of cyclohexane-based analogs (Scheme 2), thiophene-3-acetonitrile was refluxed in tert-butanol and subjected to a double Michael addition in the presence of methyl acrylate and benzyl trimethylammonium hydroxide (Triton B), yielding the diester intermediate 5. In the following step, compound 5 was deprotonated using potassium tertbutoxide and underwent Dieckmann condensation to form the 4-heteroaryl-4-cyano-2carbomethoxycyclohexanone derivative 6. The 2-carboxymethyl group was then removed by heating at 100 °C in a mixture of 10 % sulfuric acid and glacial acetic acid, producing 4-oxo-1-(thiophen-3-yl)cyclohexane-1-carbonitrile (7). The ketone group of 7 was protected with ethylene glycol, forming compound 8. Subsequent reduction of the nitrile group using LiAlH₄ in THF provided the primary amine 10, which was then acylated with various acyl chlorides to produce intermediates 12a-i. Deprotection of the ketone yielded compounds 13a-i. Selective reduction of the ketone group in 13a-i using sodium borohydride (NaBH₄) produced diastereomeric mixtures of alcohols 14a-i, which were separated into their cis- and transisomers by column chromatography. The resulting hydroxy analogs were reacted with 4nitrochloroformates to form 4-nitrophenyl carbonate intermediates, which were subsequently treated with various primary amines to yield novel carbamate derivatives *trans*-15, *trans*-16, trans-17, trans-18, trans-19, trans-20, and cis-21.

Determination of K_V1.3 inhibition

To expand the library of 2-methoxybenzamides, a series of new $K_V 1.3$ inhibitors was synthetized and evaluated for their inhibitory activity at 1 μ M and 10 μ M using whole-cell

patch-clamp electrophysiology (Tables 1-3). Replacement of the 2-methoxyphenyl moiety in compound **B** (Fig. 2) with a 2-methoxycyclohexyl, as in **11b**, resulted in loss of inhibitory activity, while compound **11a** with 3-methoxythiophen-4-yl group showed full inhibition of $K_V 1.3$ at 10 μ M (Table I). Within the ketone series, compound **13a**, featuring a 5-fluoro substituent on the 2-methoxyphenyl ring, exhibited the highest potency, achieving 49 % inhibition at 1 μ M and complete block at 10 μ M. Substitution with either a 4-methyl (**13c**) or 4-methoxy (**13f**) group on the 2-methoxyphenyl ring was better tolerated than a 5-methyl substituent (**13b**), which showed reduced activity. Strong inhibition was also observed when the 2-methoxyphenyl group was replaced with a 3-methoxythiophen-4-yl (**13d**) or a 2-ethoxyphenyl group (**13h**). In contrast, compounds **13e** (bearing a methoxycyclohexyl group) and **13g** (with a furan-3-yl substituent) only weakly inhibited K_V1.3 at 10 μ M (Table I).

Table I. $K_V 1.3$ inhibitory activity of new analogs **11a**, **11b** and **13a-h**, manually patch-clamped to determine the percentage of inhibition at 1 μ M and 10 μ M (n = 3)

Compound	Structure	Average % block at 1 μM	Average % block at 10 μM
11 a	S O O S S S	60 %	100 %
11b		0 %	22 %
13a		49 %	100 %
13b		9 %	63 %



In the hydroxy series (**14a-d,g,i**), an SAR pattern similar to that observed for the ketones (**13a-h**) was evident. Notably, the *cis*-isomers consistently exhibited more potent K_v 1.3 inhibition than their *trans* counterparts, regardless of the substitution on the 2-methoxyphenyl ring. Among the hydroxy derivatives, the most potent compounds were *cis*-

14c, cis-14d, and cis-14i, bearing 4-methyl-2-methoxyphenyl, 3-methoxythiophen-4-yl, and 2methoxythiophen-3-yl substituents, respectively. Each of these compounds achieved complete K_v 1.3 inhibition at 1 μ M (Table II). In the carbamate series (Table III), the 3-methoxypropyl analog of compound C (trans-15) also demonstrated full K_v1.3 inhibition at 1 µM. Consistent with findings from our previous study,¹⁶ the *trans*-carbamates generally exhibited stronger inhibitory activity than their *cis* counterparts. For example, *trans*-17 inhibited K_v1.3 by 87 % at 1 µM, whereas *cis*-21 showed only 21 % inhibition at the same concentration. Other potent inhibitors included the propylcarbamates trans-16 and trans-18, analogs of 14d and 14a, respectively, both of which showed complete inhibition at 1 µM (Table III). Based on this preliminary screening, the most promising compounds, cis-14c, cis-14d, cis-14i, trans-15, trans-16, trans-17, and trans-18, were selected for further evaluation in concentration-response experiments (Table IV).

Table II. $K_V 1.3$ inhibitory activity of new analogs 14a-d,g,i, manually patch-clamped to determine the percentage of inhibition at 1 μ M and 10 μ M (n = 3)

Compound	Structure	Average % block at 1 μM	Average % block at 10 µM
trans- 14a		31 %	80 %
cis-14a	OH NH S O F	84 %	100 %
trans-14b	OH S NH O	11 %	40 %

<i>cis</i> -14b	S NH	49 %	100 %
trans- 14c		0 %	32 %
<i>cis-</i> 14c	S NH	100 %	n.t.
trans-14d	OH S O O S S S S	12 %	52 %
<i>cis</i> -14d	OH S O NH S S S S S	100 %	n.t.
trans-14g		0 %	0 %
cis- 14g	S NH O O	10 %	40 %



Table III. $K_V 1.3$ inhibitory activity of new analogs **15-21**, manually patch-clamped to determine the percentage of inhibition at 1 μ M and 10 μ M (n =3). n.t. = not tested

Compd.	Structure	Average % block at 1 µM	Average % block at 10 µM
trans-15		100 %	n.t.
trans-16		100 %	n.t.



The potency of compounds *cis*-14c, *cis*-14d, and *cis*-14i was assessed by determining their IC₅₀ values for $K_V 1.3$ inhibition and compared to the reference compounds *cis*-D (IC₅₀ =

226 nM) and *trans*-**D** (IC₅₀ = 2.2 μ M). Among the tested compounds, *cis*-**14i** showed the highest potency, with an IC₅₀ of 326 nM, followed closely by *cis*-**14d** (IC₅₀ = 346 nM) and *cis*-**14c** (IC₅₀ = 505 nM). Although all three compounds displayed submicromolar IC₅₀ values, they were less potent than *cis*-**D**. The potency of compounds *trans*-**15**, *trans*-**16**, *trans*-**17**, and *trans*-**18** was compared to the parent compound *trans*-**C** (IC₅₀ = 0.23 μ M). Among these, *trans*-**18** exhibited the highest potency, with an IC₅₀ of 122 nM, followed by *trans*-**16** with an IC₅₀ of 166 nM. While both compounds were less potent than *trans*-**15** and *trans*-**17** were less potent than *trans*-**16** and *trans*-**18**, though their inhibitory potency remained within the submicromolar range. These results suggest that the 2-methoxyphenyl ring, as present in compound *trans*-**C**, represents an optimal substituent for Kv1.3 inhibition at this position. However, small modifications, such as those present in *trans*-**16** and *trans*-**18**, are well tolerated in maintaining strong inhibitory potency.









CONCLUSION

In this study, a ligand-based design approach was used for the optimization of novel thiophenebased K_v1.3 inhibitors, building on previously reported benzamide scaffold. Structural modifications focused on the 2-methoxybenzamide moiety and the core tetrahydropyran or cyclohexane ring, while retaining the 3-thienyl group, enabled the identification of several potent inhibitors. Structure–activity relationship analysis revealed that *cis*-isomers in the hydroxy series consistently exhibited stronger K_v1.3 inhibition than their *trans* counterparts, with *cis*-14i, *cis*-14d, and *cis*-14c emerging as the most effective (IC₅₀ = 326–505 nM). In the carbamate series, *trans*-isomers showed superior activity, with *trans*-18 and *trans*-16 achieving IC₅₀ values of 122 nM and 166 nM, respectively. While none of the new compounds outperformed the most potent reference inhibitors, several analogs exhibited submicromolar potency. These findings provide valuable insights into the SAR of K_v1.3 inhibitors and highlight promising candidates for further development.

Notes

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